

5

ELECTRONS IN ATOMS

A. Matching

Match each term in Column A with the correct description in Column B. Write the letter of the correct term on the line.

Column A

- _____ 1. the ways in which electrons are arranged around the nuclei of atoms
- _____ 2. the ejection of electrons by metals when light shines on them
- _____ 3. the region around the nucleus of an atom where an electron is likely to be moving
- _____ 4. An atomic orbital may describe at most two electrons.
- _____ 5. the regions within which electrons have the highest probability of being found
- _____ 6. when electrons occupy orbitals of equal energy, one electron enters each orbital until all the orbitals contain one electron.
- _____ 7. the amount of energy required to move an electron from its present energy level to the next highest one
- _____ 8. light quanta
- _____ 9. Electrons enter orbitals of lowest energy first.
- _____ 10. the modern description of the electrons in atoms

Column B

- a. Aufbau principle
- b. quantum
- c. photoelectric effect
- d. atomic orbitals
- e. Pauli exclusion principle
- f. energy level
- g. Hund's rule
- h. electron configurations
- i. quantum mechanical model
- j. photons

B. Multiple Choice

Choose the best answer and write its letter on the line.

- _____ 11. Bohr's contribution to the development of atomic structure:
 - a. was referred to as the "planar parking model."
 - b. was the discovery that electrons surround a dense nucleus.
 - c. was proposed that electrons travel in circular orbits around the nucleus.
 - d. is the quantum mechanical model.
- _____ 12. What is the total number of orbitals in the third principal energy level?
 - a. 1
 - b. 4
 - c. 9
 - d. 16

Electrons In Atoms Chapter 5 Test B Answers

N Noddings



Electrons In Atoms Chapter 5 Test B Answers:

Unveiling the Magic of Words: A Review of "**Electrons In Atoms Chapter 5 Test B Answers**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Electrons In Atoms Chapter 5 Test B Answers**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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