

Electron Energy and Light

How does light reveal the behavior of electrons in an atom?

Why?

From fireworks to stars, the color of light is useful in finding out what's in matter. The emission of light by hydrogen and other atoms has played a key role in understanding the electronic structure of atoms. Trace materials, such as evidence from a crime scene, lead in paint or mercury in drinking water, can be identified by heating or burning the materials and examining the color(s) of light given off in the form of bright-line spectra.

Model 1 – White Light



- Trace the arrows in Model 1 and shade in the table with colored pencils where appropriate.
- What happens to white light when it passes through a prism?

The different wavelengths get separated

- Why are the color labels in the table in Model 1 plural (i.e., "Reds" rather than "Red")?

many different shades of a color in a range of wavelength

- Do all colors of light travel at the same speed?

Yes

- Do all colors of light have the same energy? If no, which colors have the highest energy and the least energy, respectively?

*All colors do not have the same energy. Blues + violets highest
Reds + oranges lowest*

- Consider the light illustrated in Model 1.

- Which color corresponds to the longest wavelengths?

Reds

- Which color corresponds to the shortest wavelengths?

Violets

- Write a sentence that describes the relationship between wavelength and energy of light.

As the wavelength gets shorter the energy of light increases

Electron Energy And Light Answer Sheet

Lei Huang



Electron Energy And Light Answer Sheet:

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specific science pedagogies of senior high school biology chemistry and physics It builds a compatible alignment of multiple representation and representation construction approaches to science pedagogy with the social semiotic systemic functional linguistic based approaches to explicit teaching of disciplinary literacy The early part of the book explicates the transdisciplinary negotiated theoretical underpinning of the MDL framework followed by the research informed repertoire of learning experiences that are then articulated into a comprehensive framework of options for the planning of classroom work Practical adoption and adaptation of the framework in biology chemistry and physics classrooms are detailed in separate chapters The latter chapters indicate the impact of the collaborative research on teachers professional learning and students multimodal disciplinary literacy engagement concluding with proposals for accommodating emerging developments in MDL in an ever changing digital communication world The MDL framework is designed to enable teachers to develop all students disciplinary literacy competencies This book will be of interest to researchers teacher educators and postgraduate students in the field of science education It will also have appeal to those in literacy education and social semiotics The Open Access version of this book available at www.taylorfrancis.com has been made available under a Creative Commons Attribution Non Commercial No Derivatives 4.0 license

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