

Name: Ahmad Hamoud Date: 2/11/2020

Student Exploration: Electron Configuration

Vocabulary: atomic number, atomic radius, Aufbau principle, chemical family, diagonal rule, electron configuration, Hund's rule, orbital, Pauli exclusion principle, period, shell, spin, subshell

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Elvis Perkins, a rather shy fellow, is getting on the bus shown at right. Which seat do you think he will probably sit in? Mark this seat with an "E."
2. Marta Warren gets on the bus after Elvis. She is tired after a long day at work. Where do you think she will sit? Mark this seat with an "M."
3. In your experience, do strangers getting on a bus like to sit with other people if there is an empty seat available? NO



Gizmo Warm-up

Just like passengers getting on a bus, electrons orbit the nuclei of atoms in particular patterns. You will discover these patterns (and how electrons sometimes act like passengers boarding a bus) with the *Electron Configuration Gizmo*.

To begin, check that **Lithium** is selected on the PERIODIC TABLE tab.

1. The **atomic number** is equal to the number of protons in an atom.

How many protons are in a lithium atom? 3

2. A neutral atom has the same number of electrons and protons.

How many electrons are in a neutral lithium atom? 3

3. Select the ELECTRON CONFIGURATION tab, and check that **Energy** is selected. Click twice in the **1s** box at lower left and once in the **2s** box. Observe the atom model at right.

A. What do you see? a nucleus with somethings orbiting it



| 3 | 4 | 5 | 6 | 7 | 8 |
|----|----|----|----|----|----|
| Sc | Ti | V | Cr | Mn | Fe |
| Y | Zr | Nb | Mo | Tc | Ru |
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Table of Contents Electron Configuration Gizmo Answers

1. Understanding the eBook Electron Configuration Gizmo Answers
 - The Rise of Digital Reading Electron Configuration Gizmo Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Electron Configuration Gizmo Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Electron Configuration Gizmo Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Electron Configuration Gizmo Answers
 - Personalized Recommendations

- Electron Configuration Gizmo Answers User Reviews and Ratings
- Electron Configuration Gizmo Answers and Bestseller Lists
- 5. Accessing Electron Configuration Gizmo Answers Free and Paid eBooks
 - Electron Configuration Gizmo Answers Public Domain eBooks
 - Electron Configuration Gizmo Answers eBook Subscription Services
 - Electron Configuration Gizmo Answers Budget-Friendly Options
- 6. Navigating Electron Configuration Gizmo Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Electron Configuration Gizmo Answers Compatibility with Devices
 - Electron Configuration Gizmo Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Electron Configuration Gizmo Answers
 - Highlighting and Note-Taking Electron Configuration Gizmo Answers
 - Interactive Elements Electron Configuration Gizmo Answers
- 8. Staying Engaged with Electron Configuration Gizmo Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Electron Configuration Gizmo Answers
- 9. Balancing eBooks and Physical Books Electron Configuration Gizmo Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Electron Configuration Gizmo Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Electron Configuration Gizmo Answers
 - Setting Reading Goals Electron Configuration Gizmo Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Electron Configuration Gizmo Answers
 - Fact-Checking eBook Content of Electron Configuration Gizmo Answers

- Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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