

সুতরাং, বিচ্ছেদ MY_2 অসম্ভব হবে।

☀️ 1R1b 1vLv 1v4jv:

যদি জলসমাধানের কোনো প্রকরণ একটি লবণ প্রস্তুত করতে থাকবে যদি অসম্ভব হবে তা নির্ভর করে লবণটির প্রকরণ জলসমাধান উপর। যদি প্রকরণ কোনো লবণের উপস্থানে অসম্ভবতার আয়নিক ভারসাম্যের মান লবণটির প্রকরণ জলসমাধানের মানকে অতিক্রম করতে না পারে তবে লবণটি প্রকরণে প্রস্তুত করা যাবে। আর যদি লবণের আয়নিক ভারসাম্যের মান (K_{sp}) লবণটির প্রকরণ জলসমাধানের মান (K_{sp}) কে অতিক্রম করে তবে লবণটি প্রকরণ থেকে অসম্ভব হবে।

অর্থাৎ,

- $K_{sp} < K_{sp}$: প্রকরণ অসম্ভব প্রকরণ
- $K_{sp} = K_{sp}$: প্রকরণ সম্ভব প্রকরণ
- $K_{sp} > K_{sp}$: প্রকরণ থেকে প্রকরণ অসম্ভব হবে

III A পরে PN প্রকরণ থেকে প্রকরণ তা বিচ্ছেদে অসম্ভব হবে।



সুতরাং, PN প্রকরণের ঘনত্ব, $0.1M$ হবে-

P^{+} আয়নের ঘনত্ব, $[P^{+}] = 0.1 M$

N^{-} আয়নের ঘনত্ব, $[N^{-}] = 0.1 M$

এখানে, PN প্রকরণ N^{-} আয়ন উপস্থিত অসম্ভবিক A পরে MN_2 লবণের একটি অংশ N^{-} উপস্থিত থাকবে A পরে PN প্রকরণ থেকে প্রকরণ সম্ভবতার প্রকরণের কারণে MN_2 এর প্রকরণের কারণে।

যদি, MN_2 এর লবণ প্রকরণ, S

প্রকরণ MN_2 বিচ্ছেদে অসম্ভবিক হবে,



সুতরাং, M^{2+} আয়নের ঘনত্ব, $[M^{2+}] = S$

N^{-} আয়নের ঘনত্ব, $[N^{-}] = 3S$

সুতরাং, N^{-} আয়ন PN ও MN_2 উভয় সীমিত উপস্থিত, সুতরাং

N^{-} আয়নের মোট ঘনত্ব হবে, $[N^{-}] = (3S + 0.1)M$

এখন MN_2 এর আয়নিক ভারসাম্য,

$K_{sp} = [M^{2+}] \times [N^{-}]^2$

যদি, $2.7 \times 10^{-11} = S \times (3S + 0.1)^2$ [∵ K_{sp} হবে, $K_{sp} = 2.7 \times 10^{-11}$]
 $= S \times [(3S)^2 + 3 \times (3S) \times 0.1 + 0.1^2] = 0.1 + 3 \times 3S \times 0.1 + (0.1)^2$
 $= S \times [27S^2 + 2.7S + 0.09]$
 $= 27S^3 + 2.7S^2 + 0.09S^2 + 0.001S$

যদি, $2.7 \times 10^{-11} = 0.001S$ [∵ S^3, S^2, S^2 এর মান খুবই লক্ষ্য]

∴ $S = \frac{2.7 \times 10^{-11}}{0.001}$
 $= 2.7 \times 10^{-8} \text{ mol L}^{-1}$

অর্থাৎ, MN_2 সীমিত লবণ প্রকরণ, $2.7 \times 10^{-8} \text{ mol L}^{-1}$

সুতরাং, A পরে PN প্রকরণ থেকে প্রকরণ MN_2 এর প্রকরণ $10^{-8} \text{ mol L}^{-1}$ থেকে-প্রকরণ থেকে $2.7 \times 10^{-8} \text{ mol L}^{-1}$ হবে।



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