

(vi) What is the IUPAC name of



?

- (a) α - Amino butanoic acid
(c) α - Amino butyric acid

- (b) 2 - Amino butyric acid
(d) 2 - Amino butanoic acid

(vii) Which among the following molecular formulae represents urotropine?

- (a) $C_6H_{12}N_4$ (b) $C_6H_{24}H_4$ (c) $C_6H_{12}N_4O_2$ (d) $C_6H_{24}N_4O_2$

Q. 6. Answer any THREE of the following:

[12]

- Write the structures of –
(a) 3-chloro-3-ethylhex-1-ene
(b) 1-Iodo-2, 3-dimethylbutane
(c) 1, 3, 5 - tribromobenzene
- What is the action of acidified potassium dichromate on – (a) SO_2 (b) KI
Draw structure of dichromate ion.
- Describe laboratory method for preparation of glucose.
Write the reaction that indicates the presence of $-CHO$ group in glucose.
- What will be the action of the mixture of sodium nitrite and dilute hydrochloric acid on – (a) ethyl amine (b) aniline (c) triethyl amine

Q. 7. Answer any SIX of the following :

[9]

- What are chemical twins? Write 'two' examples.
- Explain the terms : (a) Antiseptics (b) Analgesics
- Draw the simple Fisher projection formulae of D - (+) - glucose and D - (-) - fructose.
- Classify the following ligands into monodentate and polydentate –
(a) Ammonia (b) Carbon monoxide
(c) Ethylene diamine (d) Ethylene diamine tetra acetate ion
- State and explain Markownikoff's rule with suitable example.
- How are propan-1-amine and propan-2-amine prepared from oxime?
- Identify 'A' and 'B' in the following reaction :
$$C_6H_5MgBr + CO_2 \xrightarrow[H^+/H_2O]{\text{Dry ether}} 'A' \xrightarrow{PCl_5} 'B'$$
- What is the action of the following reagents on phenol –
(a) Bromine in CS_2 at low temperature.
(b) Conc. H_2SO_4 at room temperature.

Q. 8. Answer any ONE of the following:

[7]

- Write the structure and IUPAC names of all the metamers represented by formula $C_4H_{10}O$.

Write balanced chemical equations for action of ammonia on –

- (a) formaldehyde (b) acetaldehyde (c) acetone

- Write 'four' characteristics of co-ordinate complex ion.

How is Nylon 6, 6 prepared?

Write any 'two' uses of terylene.

Write 'four' physical methods of preserving food materials.

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