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Stoichiometry and Percent Yield Practice

Answer the following questions about theoretical yield and percent yield. Make sure you show all work, units, and round for sig figs.

1. Use the following equation to answer the questions below.



If I start this reaction with 57.0 grams of nitrogen with excess hydrogen, how many grams of ammonia (NH_3) can theoretically be produced?

$$57.0 \text{ g N}_2 \times \frac{1 \text{ mol N}_2}{28.014 \text{ g N}_2} \times \frac{2 \text{ mol NH}_3}{1 \text{ mol N}_2} \times \frac{17.031 \text{ g NH}_3}{1 \text{ mol NH}_3} = \boxed{69.3 \text{ g NH}_3}$$

If 60.0 grams of ammonia is actually produced, what was my percent yield of ammonia?

$$\frac{60.0}{69.3} \times 100 = \boxed{86.6\%}$$

2. Use the following equation to answer the questions below.



If 11.3 grams of carbon dioxide gas is used in this reaction, how many grams of lithium carbonate can theoretically be produced?

$$11.3 \text{ g CO}_2 \times \frac{1 \text{ mol CO}_2}{44.009 \text{ g CO}_2} \times \frac{1 \text{ mol Li}_2\text{CO}_3}{1 \text{ mol CO}_2} \times \frac{73.87 \text{ g Li}_2\text{CO}_3}{1 \text{ mol Li}_2\text{CO}_3} = \boxed{19.0 \text{ g Li}_2\text{CO}_3}$$

If 14.4 grams of lithium carbonate are actually produced, what was my percent yield?

$$\frac{14.4}{19.0} \times 100 = \boxed{75.8\%}$$

Experiment 7 Reaction Stoichiometry Percent Yield Answers

M Carnoy



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Embracing the Tune of Phrase: An Psychological Symphony within **Experiment 7 Reaction Stoichiometry Percent Yield Answers**

In a world taken by monitors and the ceaseless chatter of instant transmission, the melodic beauty and mental symphony created by the published word usually diminish in to the backdrop, eclipsed by the relentless noise and interruptions that permeate our lives. But, nestled within the pages of **Experiment 7 Reaction Stoichiometry Percent Yield Answers** a stunning fictional treasure filled with raw feelings, lies an immersive symphony waiting to be embraced. Constructed by an elegant composer of language, that captivating masterpiece conducts viewers on an emotional journey, well unraveling the concealed songs and profound influence resonating within each cautiously crafted phrase. Within the depths of this touching analysis, we shall explore the book is main harmonies, analyze its enthralling publishing model, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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