

Option (iii) is the answer.

6. Low concentration of oxygen in the blood and tissues of people living at high altitude is due to _____.

- (i) low temperature
- (ii) low atmospheric pressure
- (iii) high atmospheric pressure
- (iv) both low temperature and high atmospheric pressure

Solution:

Option (ii) is the answer.

7. Considering the formation, breaking and strength of hydrogen bond, predict which of the following mixtures will show a positive deviation from Raoult's law?

- (i) Methanol and acetone.
- (ii) Chloroform and acetone.
- (iii) Nitric acid and water.
- (iv) Phenol and aniline.

Solution:

Option (i) is the answer.

8. Colligative properties depend on _____.

- (i) the nature of the solute particles dissolved in solution.
- (ii) the number of solute particles in solution.
- (iii) the physical properties of the solute particles dissolved in solution.
- (iv) the nature of solvent particles.

Solution:

Option (ii) is the answer.

9. Which of the following aqueous solutions should have the highest boiling point?

- (i) 1.0 M NaOH
- (ii) 1.0 M Na₂SO₄
- (iii) 1.0 M NH₄NO₃
- (iv) 1.0 M KNO₃

Solution:

Option (ii) is the answer.

10. The unit of ebullioscopic constant is _____.

- (i) K kg mol⁻¹ or K (molality)⁻¹
- (ii) mol kg K⁻¹ or K⁻¹(molality)
- (iii) kg mol⁻¹ K⁻¹ or K⁻¹(molality)⁻¹
- (iv) K mol kg⁻¹ or K (molality)

Solution:

Option (i) is the answer.

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