#### ATOMS AND PERIODIC TABLE SUIZ

Do Not Write on This Testill You may use a periodic table.

#### Form B

#### Choose the BEST answer.

- The zigzag line on the periodic table divides
- a, alkali metals and transition metals c. semimetals and transition metals
- b. metals and nonmetals d. inert gases and halogens
- The smallest particle into which an element can be divided and still be the same substance is a (n)
- a. electron b. proton c. atom d. molecule
- Electrons are particles that
- a. help make up the nucleus and have no charae
- b. help make up the nucleus and are positively charged
- are located outside of the nucleus and are negatively charged
- d. are located outside of the nucleus and have no charge
- If you found a Carbon 13 atom, you would know that
- a, it has 13 protons c. it has 13 neutrons
- b. it has 13 electrons d. it has 7 neutrons
- What particles make up the nucleus?
- a, electrons and neutrons b. electrons and protons
- c, protons and neutrons d, electrons, protons, and neutrons
- 6. Neutrons are particles that
- a. help make up the nucleus and have no charge
- b. help make up the nucleus and are positively charged
- are located outside of the nucleus and are negatively charaed
- d, are located outside of the nucleus and have no charae
- The current model of the atom suggests that
- a, proton clouds surround the nucleus
- b. electron clouds surround the nucleus
- electrons travel in definite paths around the nucleus
- d. the exact path of a moving electron can be predicted
- The mass number minus the atomic number is equal to the number of
- a, protons and neutrons b. protons
- c. protons and electrons d. neutrons
- Mendeleev arranged the elements
- a. density b. melting point c. how they look (appearance) d. atomic mass

- 10. Which of the following is **NOT** found on the periodic table?
- a. the atomic number of each element
- b. the symbol of each element

c. the density of each element

d. the atomic mass of each element

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