

SOLVED QUESTIONS

Multiple Choice Questions

1. K_1 and K_2 are the velocity constants of forward and backward reactions. The equilibrium constant k of the reaction is
- (A) $K_1 \times K_2$ (B) $K_1 - K_2$
 (C) $\frac{K_1}{K_2}$ (D) $\frac{K_1 + K_2}{K_1 - K_2}$
- Sol.** (C).
2. Which of the following reactions will be favoured at low pressure?
- (A) $N_2 + O_2 \rightleftharpoons 2NO$ (B) $H_2 + I_2 \rightleftharpoons 2HI$
 (C) $PCl_5 \rightleftharpoons 2PCl_3 + Cl_2$ (D) $N_2 + 3H_2 \rightleftharpoons 2NH_3$
- Sol.** (C).
3. For the reaction, $A + B \longrightarrow C + D$, if active mass of A is kept constant and active mass of B is tripled, the rate of reaction will become
- (A) three times (B) six times
 (C) eight times (D) nine times
- Sol.** (D).
4. The rate of $\frac{K_p}{K_c}$ for the reaction:
- $$CO(g) + \frac{1}{2}O_2(g) \rightleftharpoons CO_2(g)$$
- is
- (A) 1 (B) RT
 (C) $(RT)^{1/2}$ (D) $(RT)^{3/2}$
- Sol.** (D).
5. Consider the reaction,
- $$CaCO_3(s) \rightleftharpoons CaO(s) + CO_2(g)$$
- in a closed container at equilibrium. What would be the effect of addition of $CaCO_3$ on the equilibrium concentration of CO_2 ?
- (A) Increases (B) Unpredictable
 (C) Decreases (D) Remains unaffected
- Sol.** (D).
6. For the reaction, $PCl_5(g) \rightleftharpoons PCl_3(g) + Cl_2(g)$, the forward reaction at constant temperature is favoured by
- (A) introducing an inert gas at constant volume
 (B) introducing $PCl_3(g)$ at constant volume
 (C) introducing $PCl_5(g)$ at constant volume
 (D) introducing $Cl_2(g)$ at constant volume
- Sol.** (C).
7. A liquid is in equilibrium with its vapour at its boiling point. On an average, the molecules in the two phases have equal
- (A) intermolecular forces
 (B) potential energy
 (C) kinetic energy
 (D) both kinetic energy and potential energy
- Sol.** (C).
8. The unit of equilibrium constant for the reversible reaction

Multiple Choice Questions In Chemical Equilibrium

Marcel A. Müller



Multiple Choice Questions In Chemical Equilibrium:

Embracing the Melody of Expression: An Mental Symphony within **Multiple Choice Questions In Chemical Equilibrium**

In some sort of consumed by displays and the ceaseless chatter of instantaneous conversation, the melodic splendor and mental symphony created by the prepared term frequently diminish into the background, eclipsed by the constant noise and distractions that permeate our lives. But, located within the pages of **Multiple Choice Questions In Chemical Equilibrium** a charming fictional treasure brimming with raw thoughts, lies an immersive symphony waiting to be embraced. Crafted by a wonderful musician of language, that interesting masterpiece conducts visitors on a psychological journey, well unraveling the hidden tunes and profound influence resonating within each cautiously constructed phrase. Within the depths of the moving analysis, we will discover the book is main harmonies, analyze their enthralling writing type, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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