



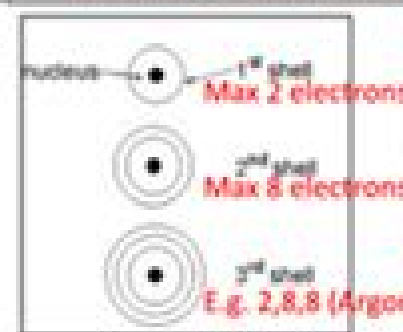
Periodic Table of the Elements

An atom:
 Number of Protons = Number of Electrons
 Mass number = Protons + Neutrons

C2 Chemistry

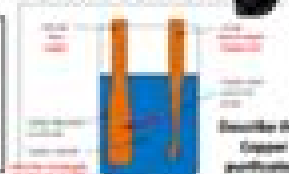
Periodic table arranges elements by **ATOMIC NUMBER** (proton number/the small one)

The number of outer shell electrons match the group the element is found in.
 E.g. Lithium 2,1 is a group 1 element.

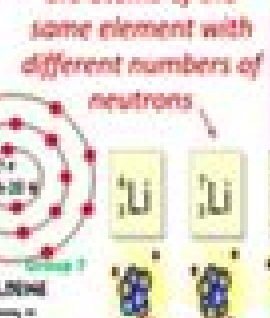
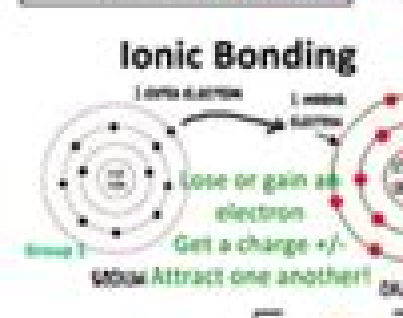
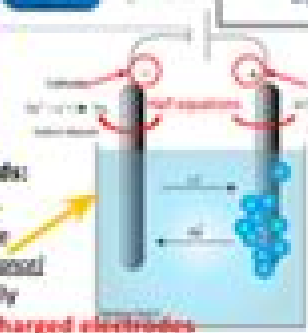


ISOTOPES
 are atoms of the same element with different numbers of neutrons

OIL RIG
 Oxidation is loss (of electrons)
 Reduction is gain (of electrons)



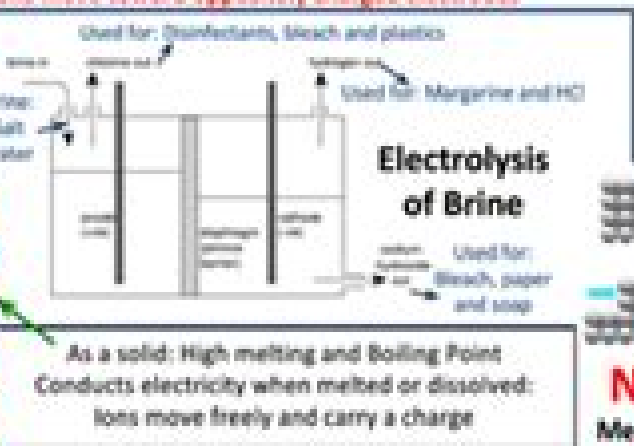
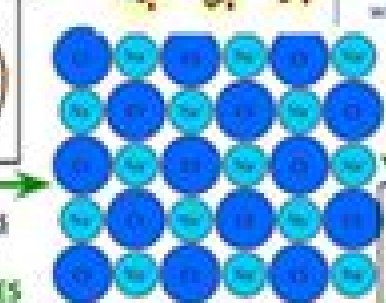
Electrolysis:
 Splitting up a substance using electricity



When there are 2 possible products
 The less reactive forms at the electrode

Electrolysis needs:
 • 2 electrodes
 • An electrolyte (ionic or molten substance)
 • A power supply

Always metals with non-metals
 + ION and - ION attracted by **STRONG ELECTROSTATIC FORCES**



Metal ions only arranged in layers
ELECTROSTATIC ATTRACTION
 between
 +IONS and ELECTRONS

Metal ions slide over one another making them **MALLEABLE**

Metals conduct **HEAT** well

NANO-TECHNOLOGY
 Means **REALLY REALLY REALLY** small!!!!

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1. Understanding the eBook Gcse Additional Chemistry C2 Revision Booklet
 - The Rise of Digital Reading Gcse Additional Chemistry C2 Revision Booklet
 - Advantages of eBooks Over Traditional Books
2. Identifying Gcse Additional Chemistry C2 Revision Booklet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gcse Additional Chemistry C2 Revision Booklet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gcse Additional Chemistry C2 Revision Booklet
 - Personalized Recommendations
 - Gcse Additional Chemistry C2 Revision Booklet User Reviews and Ratings
 - Gcse Additional Chemistry C2 Revision Booklet and Bestseller Lists
5. Accessing Gcse Additional Chemistry C2 Revision Booklet Free and Paid eBooks

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 - Gcse Additional Chemistry C2 Revision Booklet eBook Subscription Services
 - Gcse Additional Chemistry C2 Revision Booklet Budget-Friendly Options
6. Navigating Gcse Additional Chemistry C2 Revision Booklet eBook Formats
- ePub, PDF, MOBI, and More
 - Gcse Additional Chemistry C2 Revision Booklet Compatibility with Devices
 - Gcse Additional Chemistry C2 Revision Booklet Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Gcse Additional Chemistry C2 Revision Booklet
 - Highlighting and Note-Taking Gcse Additional Chemistry C2 Revision Booklet
 - Interactive Elements Gcse Additional Chemistry C2 Revision Booklet
8. Staying Engaged with Gcse Additional Chemistry C2 Revision Booklet
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gcse Additional Chemistry C2 Revision Booklet
9. Balancing eBooks and Physical Books Gcse Additional Chemistry C2 Revision Booklet
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gcse Additional Chemistry C2 Revision Booklet
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Gcse Additional Chemistry C2 Revision Booklet
- Setting Reading Goals Gcse Additional Chemistry C2 Revision Booklet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gcse Additional Chemistry C2 Revision Booklet
- Fact-Checking eBook Content of Gcse Additional Chemistry C2 Revision Booklet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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