

Form One Chemistry Exam

Duration: 1 hour 30 minutes

Instructions:

- Answer all questions.
- Show all working where applicable.
- Write your answers clearly.

SECTION A: Multiple Choice Questions (1 mark each)

- 1. What is the smallest unit of matter?
A. Atom B. Molecule C. Compound D. Element
- 2. Which one of the following is a compound?
A. Iron B. Oxygen C. Water D. Copper
- 3. Which state of matter has a definite volume but no fixed shape?
A. Solid B. Liquid C. Gas D. Plasma
- 4. The chemical symbol for Sodium is:
A. S B. N C. Na D. So
- 5. Which of the following is NOT a physical change?
A. Melting of ice B. Boiling water C. Rusting of iron D. Dissolving salt in water

SECTION B: Short Answer Questions (2 marks each)

6. Define chemistry.
7. List two differences between a physical change and a chemical change.
8. Name the three common states of matter and give one example for each.
9. What is a mixture? Give two examples.
10. What is the main difference between an element and a compound.

SECTION C: Structured Questions

11. (a) State any three laboratory safety rules. (3 marks)
(b) Why is it dangerous to taste chemicals in the lab? (2 marks)
12. (a) Name any four apparatus used in the chemistry laboratory. (2 marks)
(b) State one use for each apparatus mentioned above. (2 marks)
13. The diagram below shows the separation of a mixture of sand and salt:
(a) What method is used to separate the mixture? (1 mark)
(b) List the steps involved in the separation. (4 marks)

Form One Chemistry Revision Questions

Michael Hoelscher



Form One Chemistry Revision Questions:

This book delves into Form One Chemistry Revision Questions. Form One Chemistry Revision Questions is a crucial topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Form One Chemistry Revision Questions, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Form One Chemistry Revision Questions
- Chapter 2: Essential Elements of Form One Chemistry Revision Questions
- Chapter 3: Form One Chemistry Revision Questions in Everyday Life
- Chapter 4: Form One Chemistry Revision Questions in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Form One Chemistry Revision Questions. This chapter will explore what Form One Chemistry Revision Questions is, why Form One Chemistry Revision Questions is vital, and how to effectively learn about Form One Chemistry Revision Questions.
3. In chapter 2, the author will delve into the foundational concepts of Form One Chemistry Revision Questions. The second chapter will elucidate the essential principles that must be understood to grasp Form One Chemistry Revision Questions in its entirety.
4. In chapter 3, this book will examine the practical applications of Form One Chemistry Revision Questions in daily life. This chapter will showcase real-world examples of how Form One Chemistry Revision Questions can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Form One Chemistry Revision Questions in specific contexts. This chapter will explore how Form One Chemistry Revision Questions is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Form One Chemistry Revision Questions. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Form One Chemistry Revision Questions.

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