

NAME _____ INDEX NUMBER _____

SCHOOL _____ DATE _____

SIMPLE CLASSIFICATION OF SUBSTANCES

1. 1995 Q 15

Explain how you would separate a mixture of nitrogen and oxygen gases given that their boiling points are -196°C and -183°C respectively. (2 marks)

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2. 1995 Q 8 P2

(a) The table below gives information about the major constituents of crude oil. Study it and answer the questions that follow.

Constituent	Boiling point ($^{\circ}\text{C}$)
Gases	Below -90
Petrol	$40-175$
Kerosene	$175-250$
Diesel oil	$250-350$
Lubricating oil	$350-400$
Bitumen	Above 400

(i) Which one of the constituents of crude oil has molecules with the highest number of carbon atoms? (2 marks)

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(ii) Name the process you would use to separate a mixture of petrol and diesel and explain how the separation takes place. (2 marks)

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(iii) Explain why the constituent of crude oil and write its formula. (1 mark)

Form Chemistry Questions

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