



Name: Steven Clarke

Date: 05-02-2022

Student Exploration: Collision Theory

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: activated complex, catalyst, chemical reaction, concentration, enzyme, half-life, molecule, product, reactant, surface area

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Suppose you added a spoonful of sugar to hot water and another to ice-cold water. Which type of water will cause the sugar to dissolve more quickly?

The hot water

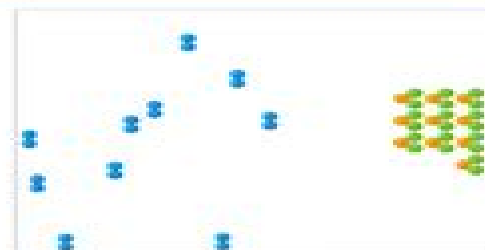
2. Suppose you held a lighted match to a solid hunk of wood and another match to a pile of wood shavings. Which form of wood will catch fire more easily?

Timber shavings

Gizmo Warm-up

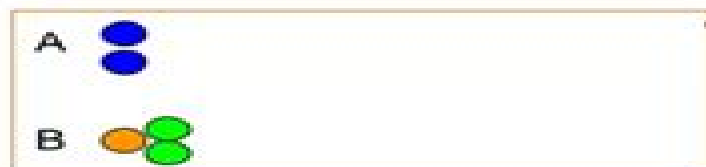
A **chemical reaction** causes the chemical compositions of substances to change. **Reactants** are substances that enter into a reaction, and **products** are substances produced by the reaction. The *Collision Theory* Gizmo allows you to experiment with several factors that affect the rate at which reactants are transformed into products in a chemical reaction.

You will need blue, green, and orange markers or colored pencils for the first part of this activity.



1. Look at the key at the bottom of the SIMULATION pane. In the space below, draw () the two reactants and two products of this chemical reaction.

Reactants:



Products:



2. Click **Play** (). What do you see?

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Albert A Gayle

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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Table of Contents Exploration Guide Collision Theory Gizmo Answer Key

1. Understanding the eBook Exploration Guide Collision Theory Gizmo Answer Key
 - The Rise of Digital Reading Exploration Guide Collision Theory Gizmo Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Exploration Guide Collision Theory Gizmo Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Exploration Guide Collision Theory Gizmo Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Exploration Guide Collision Theory Gizmo Answer Key
 - Personalized Recommendations
 - Exploration Guide Collision Theory Gizmo Answer Key User Reviews and Ratings
 - Exploration Guide Collision Theory Gizmo Answer Key and Bestseller Lists

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 - Exploration Guide Collision Theory Gizmo Answer Key Budget-Friendly Options
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 - Exploration Guide Collision Theory Gizmo Answer Key Compatibility with Devices
 - Exploration Guide Collision Theory Gizmo Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Exploration Guide Collision Theory Gizmo Answer Key
 - Highlighting and Note-Taking Exploration Guide Collision Theory Gizmo Answer Key
 - Interactive Elements Exploration Guide Collision Theory Gizmo Answer Key
8. Staying Engaged with Exploration Guide Collision Theory Gizmo Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Exploration Guide Collision Theory Gizmo Answer Key
9. Balancing eBooks and Physical Books Exploration Guide Collision Theory Gizmo Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Exploration Guide Collision Theory Gizmo Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Exploration Guide Collision Theory Gizmo Answer Key
 - Setting Reading Goals Exploration Guide Collision Theory Gizmo Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Exploration Guide Collision Theory Gizmo Answer Key
 - Fact-Checking eBook Content of Exploration Guide Collision Theory Gizmo Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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