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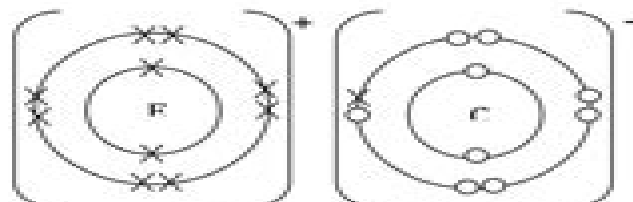
CHEMISTRY PAPER 2 (233/2)

FORM FOUR (4)

TIME: 2 HOURS

MARKING SCHEME

1. 1 (a) Alkaline-earth metals (1mk)
 (b) B (1mk)
 (c) (i) A_2B_3 (½mk) (ii) H_2SO_4 (½mk)
 (d)



Distribution of electrons (1mk) and charges (1mk)

No label (0mk)

- (e) D has a smaller atomic size than C // C has larger atomic size than D (1mk)
 D has more protons than C // D has stronger nuclear charge than C (1mk)
 (f) H (½mk). Loses electrons most readily / largest atomic radius / weakest nuclear charge (explanation ½mk)
 (g) $2A_{(s)} + 6HCl_{(aq)} \longrightarrow 2ACl_{3(aq)} + 3H_{2(g)}$ 1mk
 Moles of A = $\frac{0.081}{27} = 0.003$ moles ½mk
 Mole ratio 1:3
 Moles of HCl = $0.003 \times 3 = 0.009$ moles ½mk
 $\frac{0.009 \times 1000}{20}$ ½mk = 0.45M ½mk
 (h) The melting point increase with increase in atomic number. 1mk

This is due to increase in the strength of van der waal's forces (inter-molecular forces) 1mk

Marking Scheme For Chemistry Paper2 2002

S Ben Porath



Marking Scheme For Chemistry Paper2 2002:

Marking Scheme For Chemistry Paper2 2002 Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Marking Scheme For Chemistry Paper2 2002**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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