WORD EQUATIONS

Name <u>Key</u>

write the world equations below as chemical equations and balance.

1. zinc + lead (ii) nitrate yield zinc nitrate + lead

 $Z_{1} + P6(NO_{3})_{2} \rightarrow Z_{1}(NO_{3})_{2} + P6$

2. aluminum bromide + chlorine yield aluminum chloride + bromine

2A1Br, +3C/2 →2A1C/3+3Br2

 sodium phosphate + coloium chloride yield calcium phosphate + sodium chloride

 $ZNL_{1}, PO_{V(ng)} + 3CaCl_{2(ng)} \longrightarrow Ca_{1}(Ry)_{2(n)} + 6NhCl_{ng}$

4. potassium chiarate when heated yields potassium chiaride + axygen gas

2 KC103 - →2KC1 +302

5. aluminum + hydrochloric acid yield aluminum chloride + hydrogen gas

2A1 +6HC1 -> 2A1C1, +3H2

6. calcium hydroxide + phasphoric acid yield calcium phasphate + water

3 Ca (OH)_{2 (mg)} +2H, PO_{4 (g)} → Ca, (PO₄)_{2g} +6H₂O_g 5/44+4 506666

7. copper + sulfuric acid yield copper (ii) sulfate + water + sulfur dioxide

Cu + 2H2504 -> Cu504+2H20 + 502

hydrogen + nitrogen monoxide yield water + nitrogen

 $2H_2 + 2NO \rightarrow 2H_2O + N_3$

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Daryl Vriesenga,Instructional Fair (Firm)

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