

ক) ভারী হ্যালাজেনের সাথে বিজোড় মণ্ডথ্যক হালকা হ্যালাজেন যুক্ত হয়ে আন্তঃহ্যালাজেন যৌগ গঠন করে।

৬ ছায়া শ্যালোজেন: শ্যালোজেন নয় কিন্তু শ্যালোজেনের মতোই রাসায়নিক বিক্রিয়া দেয়। যেনন: CN^- , CNS^- ,

ক) ট্রাম-ইউরেনিয়াম মৌল: (৭৩-১০৩) পারমাণবিক
 সংখ্যা বিশিষ্ট মৌলগুলোকে ট্রাম-ইউরেনিয়াম মৌল
 বলে।

ক) কৃত্রিম মৌল: (৩৩-১১৪) পারমাণবিক মধ্যখ্যাবিমিশ্র মৌলগুলোকে কৃত্রিম মৌল বলে।

ক) প্রকৃতিতে পাওয়া মৌল ৩০ টি।

উ(১-৩২) পারমানবিক সংখ্যা বিধিষ্ট মৌলগুণের মধ্যে দুটি মৌল প্রকৃতিতে পাওয়া যায় না। তা(৪৩) এক্ষেপম (৬১)।

৬. ১ ব্রক মোলের বহিঃস্থ ইলেকট্রন কাঠামো $\rightarrow ns^{1-2}$

0 20 40 60 80 100 120 140 160 180 200 220 240 260 280 300 320 340 360 380 400 420 440 460 480 500 520 540 560 580 600 620 640 660 680 700 720 740 760 780 800 820 840 860 880 900 920 940 960 980 1000

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$$\begin{aligned} \mathcal{O}(\mathbb{R}^n) &\rightarrow [\mathcal{O}] \otimes \mathcal{O}^* \\ \mathcal{O} &\rightarrow [\mathcal{O}] \otimes \mathcal{O}^* \end{aligned}$$

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$$\text{Ge}^{IV}(\text{OH})_2 \longrightarrow [\text{Ge}] \text{O}_2^-$$

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