Date Per

#### WORKSHEET: Chemical Bonding - Ionic & Covalent!

#### REMEMBER...

Name

 Ionic Bond
 between a Metal and Non-Metal
 (M + NM)

 Covalent Bond
 between a Non-Metal and Non-Metal
 (NM + NM)

PART 1: Determine if the elements in the following compounds are metals or non-metals. Describe the type of bonding that occurs in the compound.

Compound	Element 1 (metal or non-metal?)	Element 2 (metal or non-metal?)	Bond Type
NO <sub>2</sub>	N = non-metal	O = non-metal	covalent
NaCl			
SO <sub>2</sub>			
PI <sub>3</sub>			
MgBr <sub>2</sub>			
CaO			
H <sub>2</sub> O			
K <sub>2</sub> O			
AIF <sub>3</sub>			
O <sub>2</sub>			
CuCl <sub>2</sub>			
NO <sub>2</sub>			
CO <sub>2</sub>			
HF			
Rb₂S			
NBr <sub>3</sub>			
Fe <sub>2</sub> O <sub>3</sub>			
CCI4			

# **Forming Chemical Bonds Answer Key**

**Angie Papple Johnston** 

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