

NAME _____ INDEX NUMBER _____

SCHOOL _____ DATE _____

SIMPLE CLASSIFICATION OF SUBSTANCES

1. 1995 Q 15

Explain how you would separate a mixture of nitrogen and oxygen gases given that their boiling points are -196°C and -183°C respectively. (2 marks)

.....

.....

.....

2. 1995 Q 8 P2

(a) The table below gives information about the major constituents of crude oil. Study it and answer the questions that follow.

Constituent	Boiling point ($^{\circ}\text{C}$)
Gases	Below -90
Petrol	$40-175$
Kerosene	$175-250$
Diesel oil	$250-350$
Lubricating oil	$350-400$
Bitumen	Above 400

(i) Which one of the constituents of crude oil has molecules with the highest number of carbon atoms? (2 marks)

.....

.....

(ii) Name the process you would use to separate a mixture of petrol and diesel and explain how the separation takes place. (2 marks)

.....

.....

(iii) Explain why the constituent of crude oil and write its formula. (1 mark)

Form 1 Chemistry Questions

Rachel S Tattersall



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