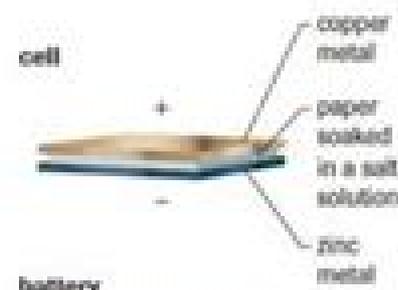


# Introduction to Electrochemistry

- ▶ An *electric cell* converts chemical energy into electrical energy
  - ▶ **Alessandro Volta** invented the first electric cell but got his inspiration from **Luigi Galvani**. Galvani's crucial observation was that two different metals could make the muscles of a frog's legs twitch. Unfortunately, Galvani thought this was due to some mysterious "animal electricity". It was Volta who recognized this experiment's potential.
  - ▶ An electric cell produces very little electricity, so Volta came up with a better design:
- ▶ A *battery* is defined as two or more electric cells connected in series to produce a steady flow of current
  - ▶ Volta's first battery consisted of several bowls of brine ( $\text{NaCl}_{(aq)}$ ) connected by metals that dipped from one bowl to another
  - ▶ His revised design, consisted of a sandwich of two metals separated by paper soaked in salt water.



# Experiment 2 electrochemical Cells Post Lab Answers

**David Kirk**



## **Experiment 2electrochemical Cells Post Lab Answers:**

Laboratory Experiments for General Chemistry Harold R. Hunt,Toby F. Block,1994     General Chemistry  
Hunt,1976-06-07     *NASA Technical Memorandum* ,1994     *Scientific and Technical Aerospace Reports* ,1995-05  
ERDA Energy Research Abstracts United States. Energy Research and Development Administration,1976     U.S.  
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Post Treatment that was published in Water     **Energy Research Abstracts** ,1992-02     Chemistry, an Experimental  
Science Chemical Education Material Study,1963     **Technical Abstract Bulletin** Defense Documentation Center  
(U.S.),1964

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