## PAPER-T

## CHEMISTRY

## (SET BY NARAYANA EDUCATIONAL INSTITUTIONS LECTURERS)

1. The minimum angular momentum of an electron with the magnetic quantum numbers (-1, 0, +1)

(1)  $\frac{\sqrt{3}h}{\pi}$  (2)  $\frac{h}{\pi}$  (3)  $\frac{2}{h\pi}$  (4)  $\frac{3h}{\pi}$ 

Arrange the following molecules in the order of increasing bond angles H,S, NH<sub>a</sub>, CH<sub>a</sub>

NH<sub>s</sub> < H<sub>s</sub>S < CH<sub>s</sub>

(2) CH<sub>4</sub> < H<sub>2</sub>S < NH<sub>3</sub>

(3) H<sub>a</sub>S < NH<sub>a</sub> < CH<sub>a</sub>

, (4) NH<sub>3</sub> < CH<sub>4</sub> < H<sub>5</sub>S

Total kinetic energy of two moles of an Ideal Gas at 127°C is

(1) 8.31 kJ

(2) 9.98 kJ

(3) 3.20 kJ

(4) 5 68 kJ

Acetic acid undergoes diments alon in Benzene. Molecular weight pol accilion Previous 30.30.20/Estion Papers 26.97% acid in Benzene was observed as 90 in an experiment. Extent of association of acetic acid in the given solvent at the given condition is

(1) 66% (2) 33% (3) 90% (4) 100%

Ammonia can be dried over

(1) Quick lime (3) P.O.

(2) Slaked lime (4) Conc. H<sub>3</sub>SO<sub>4</sub>

CH, CN + H,O → X + NH, The density of 2M solution of X in water is 1.017 gm/ml. The molality of the solution is

(2) 1.55 (3) 2.26 (4) 1.25

Electrode potentials of various metals are given  $Fe \rightarrow Fe^{+2} + 2e^{-}$ ;  $E^{0} = + 0.44V$ 

 $Pb \rightarrow Pb^{+2} + 2e^{i0}$ ;  $E^{0} = + 0.13V$  $Ag^+ + e^{\otimes} \rightarrow Ag : E^0 = + 0.8V$ 

Cu+2 + 2e<sup>50</sup> → Cu: E<sup>0</sup> = + 0.34V

Which of the following oxidise Iron?

(1) Ag only

(2) Ag, Cu only

(3) Ag, Cu, Pb (4) Pb only

Ethyl alcohol is treated with chlorine. 8. The hybridisation of carbon atoms in final product is

(1) SP3, SP3

(2) SP3 SP2

(3) SP3, SP (4) SP2

9. α-chlorination of Acetic acid in presence of phosphorous is known as

(1) Schmidt reaction

(2) Hell-vohlard-Zelensky reaction

(3) Rosenmund Reaction

(4) Sandmeyer reaction

10. Fe+2 is attached to how many nitrogen in haemoglobin

(2) 5 (3) 6

The percentage of carbon in CO2 is

12. Formula of Deadburnt Gypsum is

(1) CaSO, 2H,O (2) CaSO,

(3) 2CaSO, H,O (4) 2CaSO, 3H,O

13. In Castner Kellner Process, the Cathode in the middle compartment

(1) Hg

(2) Graphite

(3) Iron

(4) Platinum

The O-O bond is not present in

(1) H<sub>o</sub>SO<sub>c</sub> (2) CrO<sub>c</sub>

(3) HCIO, (4) HNO,

 A mixture of CH<sub>4</sub> and NH<sub>3</sub> heated to 1000°C in the presence of Al,O, can gg il was

(1) CH<sub>3</sub>NH<sub>2</sub> (2) HCN

(3) (CH<sub>2</sub>)<sub>2</sub>NH (4) CH<sub>2</sub>CN

- In carius mothod halogens are estimated as (X = halogen)

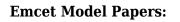
(1) X<sub>2</sub>

(2) BaX.



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#### **Table of Contents Emcet Model Papers**

- 1. Understanding the eBook Emcet Model Papers
  - The Rise of Digital Reading Emcet Model Papers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Emcet Model Papers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Emcet Model Papers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Emcet Model Papers
  - Personalized Recommendations
  - Emcet Model Papers User Reviews and Ratings
  - Emcet Model Papers and Bestseller Lists
- 5. Accessing Emcet Model Papers Free and Paid eBooks
  - Emcet Model Papers Public Domain eBooks
  - Emcet Model Papers eBook Subscription Services
  - Emcet Model Papers Budget-Friendly Options

- 6. Navigating Emcet Model Papers eBook Formats
  - o ePub, PDF, MOBI, and More
  - Emcet Model Papers Compatibility with Devices
  - Emcet Model Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Emcet Model Papers
  - Highlighting and Note-Taking Emcet Model Papers
  - Interactive Elements Emcet Model Papers
- 8. Staying Engaged with Emcet Model Papers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Emcet Model Papers
- 9. Balancing eBooks and Physical Books Emcet Model Papers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Emcet Model Papers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Emcet Model Papers
  - Setting Reading Goals Emcet Model Papers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Emcet Model Papers
  - Fact-Checking eBook Content of Emcet Model Papers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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