

Final Exam

1. Name: _____

Instructions: This Final Exam consists of 10 questions. Questions 1-5 are multiple-choice questions. Questions 6-10 are short-answer questions. Questions 1-5 are worth 2 points each. Questions 6-10 are worth 4 points each. The total score is 30 points. You must answer all questions. This is a closed-book exam. You may use a calculator.

1. The following information is given for the following reaction:
$$2\text{H}_2\text{O}(\text{l}) \rightarrow 2\text{H}_2(\text{g}) + \text{O}_2(\text{g})$$

The standard enthalpy of formation of $\text{H}_2\text{O}(\text{l})$ is -285.8 kJ/mol . The standard enthalpy of formation of $\text{H}_2(\text{g})$ is 0 kJ/mol . The standard enthalpy of formation of $\text{O}_2(\text{g})$ is 0 kJ/mol .
What is the standard enthalpy of formation of $\text{H}_2\text{O}(\text{l})$?
a. -285.8 kJ/mol
b. -571.6 kJ/mol
c. -142.9 kJ/mol
d. -117.9 kJ/mol
2. The standard enthalpy of formation of $\text{H}_2\text{O}(\text{l})$ is -285.8 kJ/mol . The standard enthalpy of formation of $\text{H}_2(\text{g})$ is 0 kJ/mol . The standard enthalpy of formation of $\text{O}_2(\text{g})$ is 0 kJ/mol .
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Microeconomics Released Multiple Choice 20answer Key

**Brian Reddington, Thomas A. editor -
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Multiple Choice Questions for Microeconomics and Macroeconomics Philip Mayer, Kenneth P. Gillian, 1999-01-01

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