



Element Builder SE ...



3	Na	Mg	3	4	5	6	7	8	9	10	11	12	Al	Si	P	S	Cl	Ar
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Why do you think the elements are arranged the way that they are? Based on how many shells and metals, non metals

2. Draw diagrams: Create an electron dot diagram for each of the elements below. Use the Gizmo to help you do this. To check your work, turn on Show electron dot diagram.

H .

He . .

Li .

. Be .

. B .

. C .

N

O

F

Ne

Na

Mg

Al

Si

P

S

Cl

Ar

3. Analyze: What do the elements in each column of the periodic table have in common?

Elements in each column have the same valence shell electron configurations, and the elements have some similar chemical properties

Draw conclusions: How is the periodic table organized? Periodic Table is based on periodic law which states that elements are arranged in order of increasing atomic numbers, then their properties are repeated in periodic manner. It is arranged in order of increasing atomic numbers

4.

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Key For Elementbuilderse

Robert R. Redfield



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