

## Assessment

## Quiz

**Section: Describing Chemical Reactions**

In the space provided, write the letter of the term or phrase that best answers the question.

- \_\_\_\_\_ 1. The formation of a gas is evidence of  
**a.** a chemical change.  
**b.** a physical change.  
**c.** no change in energy.  
**d.** Both (a) and (c).
- \_\_\_\_\_ 2. Which of the following may be evidence of a chemical change?  
**a.** release of energy as heat  
**b.** production of sound  
**c.** reduction in temperature  
**d.** All of the above
- \_\_\_\_\_ 3. In the word equation,  
iron + copper(II) sulfate  $\rightarrow$  iron(II) sulfate + copper,  
a product is  
**a.** iron(II) sulfate,  
**b.** copper,  
**c.** iron,  
**d.** Both (a) and (b)
- \_\_\_\_\_ 4. In the reaction described by the following equation  
water + energy  $\rightarrow$  hydrogen gas + oxygen gas,  
energy  
**a.** can be considered a reactant.  
**b.** is absorbed.  
**c.** is released.  
**d.** Both (a) and (b)
- \_\_\_\_\_ 5. In the unbalanced formula equation  
carbon + hydrogen  $\rightarrow$  methane, the formula  
 $\text{CH}_4$  represents  
**a.** one atom of methane.  
**b.** five atoms of methane.  
**c.** one molecule of methane.  
**d.** one gram of methane.

# Holt Chemistry Chapter Test Answers

**Sabine Zange**



## Holt Chemistry Chapter Test Answers:

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