

Atomic History Worksheet Answer Key

Democritus Proposes the Atom

1. The word *atom* comes from a Greek word that means **indivisible**.
2. Which of the following statements is part of Democritus's theory about atoms?
 - a. **Atoms are the smallest piece of matter.**
 - d. **Atoms exist but will never be fully proven.**
3. We know that Democritus was right about atoms. So why did people ignore Democritus's ideas for such a long time?
Democritus's ideas were ignored for a long time because he had no proof of the atoms existence and people believed Aristotle's ideas.

Dalton Creates an Atomic Theory Based on Experiments

4. By conducting experiments and making observations, Dalton figured out that elements combine in random proportions because they're made of individual atoms. **False**
5. Dalton's theory states that atoms cannot be **changed, created, or destroyed**.
6. Atoms of different elements are exactly alike. **False**
7. How did Dalton think atoms formed new substances?
Atoms can be combined in set proportions to form new substances.

Thomson Finds Electrons in the Atom

Mark the following statements *True* or *False*.

8. In 1897, J. J. Thomson made a discovery that proved Dalton's atomic model was correct. **False**
9. Thomson discovered that there were small particles inside the atom. **True**
10. Thomson found that the electrically charged plates affected the direction of a cathode-ray tube beam. **True**
11. Thomson knew the beam was made of particles with a positive charge because it was pulled toward a positive charge. **False**
12. The two types of charges are positive and neutral. **False**

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Record University of North Carolina (1793-1962), 1904

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Table of Contents History Of The Atom Answer Key

1. Understanding the eBook History Of The Atom Answer Key
 - The Rise of Digital Reading History Of The Atom Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying History Of The Atom Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an History Of The Atom Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from History Of The Atom Answer Key

- Personalized Recommendations
 - History Of The Atom Answer Key User Reviews and Ratings
 - History Of The Atom Answer Key and Bestseller Lists
5. Accessing History Of The Atom Answer Key Free and Paid eBooks
 - History Of The Atom Answer Key Public Domain eBooks
 - History Of The Atom Answer Key eBook Subscription Services
 - History Of The Atom Answer Key Budget-Friendly Options
 6. Navigating History Of The Atom Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - History Of The Atom Answer Key Compatibility with Devices
 - History Of The Atom Answer Key Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of History Of The Atom Answer Key
 - Highlighting and Note-Taking History Of The Atom Answer Key
 - Interactive Elements History Of The Atom Answer Key
 8. Staying Engaged with History Of The Atom Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers History Of The Atom Answer Key
 9. Balancing eBooks and Physical Books History Of The Atom Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection History Of The Atom Answer Key
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine History Of The Atom Answer Key
 - Setting Reading Goals History Of The Atom Answer Key
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of History Of The Atom Answer Key

- Fact-Checking eBook Content of History Of The Atom Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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