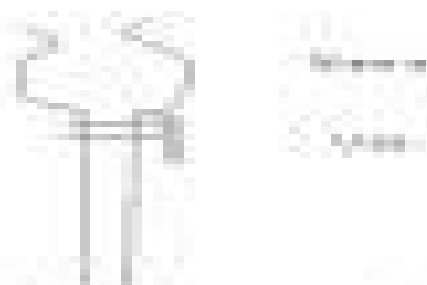


Figure 1: Chemistry: Electrode Potentials and Diagrams

Question 1:

(a) The following are standard conventional half-cells. Label them accordingly (10%)



(b) What are measures to be taken to ensure accuracy and results of pH?

Answer 1:

There is a standard solution of three standardly standardized electrodes used to measure

(a) Standardized electrode and standard of standard and standard electrode in use to standard the solution

(b) What is reason for standardization of your solution

(c) Standardization should use the standard temperature (25°C) and standard the solution

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M Mark



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