



CHEM FAX!

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Preparing and Diluting Solutions

Student Laboratory Kit

Introduction

Solutions are homogeneous mixtures. The most common solutions are aqueous solutions, which are solutions in which water is the solvent.

Concepts

- 1. Solvent
- 2. Solute

Background

Preparing Solutions

The amount of solute that is dissolved in a solution is called the concentration. The concentration of a solution can be expressed in many ways. The most common way is to express it as a percentage, which is the ratio of the mass of the solute to the mass of the solution, multiplied by 100.

When a solution is prepared, the amount of solute that is dissolved is called the mass of the solute. The amount of solvent that is used is called the mass of the solvent. The total mass of the solution is the sum of the mass of the solute and the mass of the solvent.

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Flinn Scientific Preparing And Diluting Answers**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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