It's Elementary — March Madness F

FLINN SCIENTIFIC CHEM/FAXI

March Madness is upon us? Your students are probably already working on their next assignment in bracketology, predicting the field for the NCAA Basketball Tournament. (Although the word down't appear in Webster's, there is an article on bracketology on Wikipedia, so it must be real.)

Let's apply the principles of bracketology to the properties of the elements. The competition among the 64 elements is stiff, and there is a wealth of information for students to research, in this "elementary" version of March Madness.

Review of Element Properties

Define each of the following properties of the elements and briefly describe any periodic trend in the property across a row or down a column in the periodic table.

· Tomization energy

· Amounic realities

Tournament Rules

Here are the rules for predicting the winners in each round of It's Elementary—Manch Madnes. In the event of a tie in the properties of two competing elements in any round of the tournament, the element with the larger atomic mass always wins.

- First round: Research the date of discovery of each element. In each bracket, the element that was discovered earlier
 (in its free element form) when and proceeds to the second round. If an element has been known since ancient
 times, assign it a discovery date of zero.
- Second round: Compare the ionization energy of the elements in each bracket. The element with the higher-ionization energy is the winner and advances to the Sweet 16.
- Third round (Sweet 16): Compare the group numbers of the elements—the winner is the element with the larger group number value the international (IUPAC) votem (Groups 1–18).
- Fourth round: The element with the larger atomic radius wine this round and corns a trip to the Final Four. Use
 the stornic or covalent radius only, not the van der Wash radius.
- Semifinals (Final Four): Solve the following riddles to determine the two elements that will compete for the championship.
- This "salt-malor" is also a rainuralor when its silver salt is seattered into clouds.
- Once a sedative and cure for nervous rension, the ion of this element is now a trite or commonstace expression.
- Finals: It's often said that there's no ______ in team, but it is the winner in this field.

NGSS Alignment

This laboratory activity relates to the following Nest Generation Science Standards (2013):

Disciplinary Core Ideas: Middle School MS-PSI Mauer and hs Interactions PSI A: Structure and Properties of Matter Disciplinary Core Ideas: High School

HS-PSI Mauer and his Interactions PSI A: Structure and Properties of Matter Science and Engineering Practices
Asking questions and defining problems
Analysing and interpreting day

Crosscutting Concepts Patterns Structure and function

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