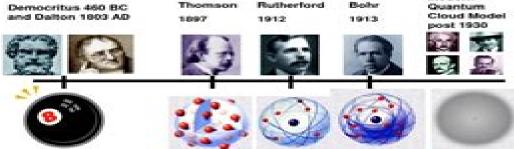
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## History of the Atom Timeline





#### John Dalton (1766 - 1844):

John Dalton was an English chemist. His ideas form the atomic theory of matter. Here are his ideas.

- All elements are composed (made up) of atoms. It is impossible to divide or destroy an atom.
- All atoms of the same elements are alike. (One atom of oxygen is like another atom of oxygen.)
- Atomic Theory

Dalton's

Modern.

- Atoms of different elements are different. (An atom of oxygen is different from an atom of hydrogen.)
- Atoms of different elements combine to form a compound. These atoms have to be in definite whole number ratios.
  For example, water is a compound made up of 2 atoms of hydrogen and 1 atom of oxygen (a ratio of 2:1). Three atoms of hydrogen and 2 atoms of oxygen cannot combine to make water.
- 1. What is the name of John Dalton's theory? THE ATOMIC THEORY OF MATTER
- 2. What are elements made of? ELEMENTS ARE MADE OF ATOMS
- 3. An atom of hydrogen and an atom of carbon are DIFFERENT
- 4. What are compounds made of? COMPOUNDS ARE MADE OF ATOMS OF TWO OR MORE DIFFERENT ELEMENTS
- 5. The ratio of atoms in HCl is: a) 1:3 b) 2:1 c) 1:1
- 6. The ratio of atoms in H<sub>2</sub>O is: a) 1:3 b) 2:1 c) 1:1 2 atoms of hydrogen and 1 atom of oxygen

# **History Of The Atom Worksheet Answers**

**Emilie Sanchez** 

**History Of The Atom Worksheet Answers:** 

Unveiling the Magic of Words: A Overview of "History Of The Atom Worksheet Answers"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**History Of The Atom Worksheet Answers**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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