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ELECTROCHEMISTRY

Study Guide

A. Matching

Match each term in Column B to the correct description in Column A.

Column A

Column B

- H
D
G
E
1. a conductor in a circuit that carries electrons to or from a substance other than a metal
 2. one part of a voltaic cell, in which either oxidation or reduction occurs
 3. a voltaic cell in which the electrolyte is a paste
 4. the electrode at which reduction occurs
 5. a voltaic cell in which a fuel undergoes oxidation and from which electrical energy is obtained continuously
 6. an electrochemical cell that is used to convert chemical energy into electrical energy
 7. a tube containing a conducting solution
 8. a group of cells that are connected together
 9. the process in which electrical energy is used to bring about a chemical change
 10. the electrode at which oxidation occurs

1. anode
1. electrode
1. cathode
1. dry cell
1. battery
1. electrolysis
1. voltaic cell
1. half-cell
1. salt bridge
1. fuel cell

B. Multiple Choice

Choose the best answer and write its letter on the line.

- D
11. In the electrolysis of brine, the substance produced at the cathode is
 - a. chlorine
 - b. oxygen
 - c. sodium
 - d. hydrogen
 12. Which of the following describes a dry cell?
 - a. It can be recharged many times.
 - b. The graphite rod does not undergo reduction, even though it is the cathode.
 - c. It contains concentrated sulfuric acid.
 - d. all of the above

High School Electrochemistry Study Guide Answers

J Spring



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