

## F324-unofficial mark scheme (in progress)

# Marks accounted for: 60/60

- Give the systematic name of the alcohol that is formed by this hydrolysis. (1)
  - Propan(e)-1,2,3-triol
- What are the other monomers that make up this polymer? (3)  
(Hydrolysed by hot NaOH)
  - Two different carboxylic acids with  $\text{COO}^-(\text{Na}^+)$  - The monomers have  $\text{COOH}$  as it's not referring to the hydrolysed products but the monomer units of the polymer I'm pretty sure that the question said: "What are the other products of this hydrolysis" or something along those lines
  - One salt had a  $\text{C}=\text{C}$  double bond
- Why are people who drink a lot of goat's milk likely to suffer from CHD? (2)
  - Fatty acids build up in the arteries and lead to CHD?
  - Has trans isomer due to  $\text{C}=\text{C}$  double bond fatty acid linked to LDL's and hence Atherosclerosis
  - Trans-fatty acid in triglyceride of milk increases 'bad' cholesterol levels in blood, resulting in increased risk of coronary heart disease and stroke (straight from the spec)
  - trans fatty acids can pack more tightly and hence have higher melting points than cis fatty acids.  $\rightarrow$  so more likely to deposit on artery walls
  - The other two molecules were fully saturated fatty acids which would also lead to the formation of atheroma
- State how 4-aminophenol can behave as a base. (1)
- The lone pair on the Nitrogen atom in the amine group can accept a proton, forming a dative covalent bond with it to give  $\text{NH}_3^+$ .
- $\text{NH}_2 + \text{H}^+ \rightarrow \text{NH}_3^+$
- 4-Aminophenol is produced by the reduction of 4-nitrophenol. Write an equation to show the production of 4-aminophenol from 4-nitrophenol. Use  $[\text{H}]$  to represent the reducing agent. (1)
- $4\text{-Nitrophenol} + 6[\text{H}] = 4\text{-Aminophenol} + 2\text{H}_2\text{O}$
- 4-nitrophenol can be produced from 4-bromophenol. (4)
  - Complete the mechanism for this reaction.
  - Use  $^+\text{NO}_2$  as the electrophile. Include any intermediate and the products.
  - In the mechanism for this reaction,  $\text{NO}_2$  substitutes for Br on the ring.

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**Rachel S Tattersall**



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