

**5 (d)** Explain why the current in the external circuit of this cell falls to zero after the cell has operated for some time.

.....

.....

*(1 mark)*

**5 (e)** The simplified electrode reactions in a rechargeable lithium cell are

Electrode **A**       $\text{Li}^+ + \text{MnO}_2 + \text{e}^- \longrightarrow \text{LiMnO}_2$        $E = -0.15 \text{ V}$

Electrode **B**               $\text{Li}^+ + \text{e}^- \longrightarrow \text{Li}$

Electrode **B** is the negative electrode.

**5 (e) (i)** The e.m.f. of this cell is 2.90 V.

Use this information to calculate a value for the electrode potential of electrode **B**.

.....

.....

*(1 mark)*

**5 (e) (ii)** Write an equation for the overall reaction that occurs when this lithium cell is being recharged.

.....

.....

.....

*(2 marks)*

**5 (e) (iii)** Suggest why the recharging of a lithium cell may lead to release of carbon dioxide into the atmosphere.

.....

.....

*(1 mark)*

# Gcse Chemistry Aqa Unit 5 June 2013

**Liying Dong**



**Gcse Chemistry Aqa Unit 5 June 2013:**

## Decoding **Gcse Chemistry Aqa Unit 5 June 2013**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Gcse Chemistry Aqa Unit 5 June 2013**," a mesmerizing literary creation penned by a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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