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Question 7

(a) In the extraction of zinc from zinc blende:

[2]

- (i) Give an equation to show how zinc oxide is converted to zinc.
- (ii) How is impure zinc finally electro-refined?
- (b) Explain why:

[3]

- (i) Transition elements form coloured compounds.
- (ii) Interhalogen compounds are more reactive than their constituent elements.
- (iii) Cu⁺ is diamagnetic but Cu²⁺ is paramagnetic. (Z = 29)

SECTION C

Answer any two questions.

Question 8

(iiii)

- (a) How can the following conversions be brought about:
 - (i) Nitro benzene to benzene diazoniumchloride.

[3]

(ii) Propanoic acid to ethylamine.

[2]

(b) Identify the compounds A, B, C, D, E and F:

Benzoic acid to benzaldehyde.

[3]

$$HC = CH \xrightarrow{diLH_2SO_4} [A] \xrightarrow{[O]} [B] \xrightarrow{SOCI_2} [C] \xrightarrow{CH_3COONa} [D]$$

$$\downarrow NH_2OH \qquad \qquad \downarrow C_6H_5OH$$

$$[F]$$

Ouestion 9

(a) Write balanced chemical equations for the following reactions and name the reactions:

[3]

- (i) Acetamide is heated with bromine and sodium hydroxide solution.
- (ii) Benzaldehyde is treated with 50% sodium hydroxide solution.
- (b) Give one chemical test to distinguish between the following pairs of compounds:

[3]

- (i) Acetone and phenol.
- (ii) Formic acid and Acetic acid.

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Léo Pomar

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- 1. Understanding the eBook Form 4 Chemistry Question Paper 2013
 - The Rise of Digital Reading Form 4 Chemistry Question Paper 2013
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Form 4 Chemistry Question Paper 2013
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Form 4 Chemistry Question Paper 2013
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Form 4 Chemistry Question Paper 2013
 - Personalized Recommendations
 - Form 4 Chemistry Question Paper 2013 User Reviews and Ratings

- Form 4 Chemistry Question Paper 2013 and Bestseller Lists
- 5. Accessing Form 4 Chemistry Question Paper 2013 Free and Paid eBooks
 - Form 4 Chemistry Question Paper 2013 Public Domain eBooks
 - Form 4 Chemistry Question Paper 2013 eBook Subscription Services
 - Form 4 Chemistry Question Paper 2013 Budget-Friendly Options
- 6. Navigating Form 4 Chemistry Question Paper 2013 eBook Formats
 - o ePub, PDF, MOBI, and More
 - Form 4 Chemistry Question Paper 2013 Compatibility with Devices
 - Form 4 Chemistry Question Paper 2013 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Form 4 Chemistry Question Paper 2013
 - Highlighting and Note-Taking Form 4 Chemistry Question Paper 2013
 - Interactive Elements Form 4 Chemistry Question Paper 2013
- 8. Staying Engaged with Form 4 Chemistry Question Paper 2013
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Form 4 Chemistry Question Paper 2013
- 9. Balancing eBooks and Physical Books Form 4 Chemistry Question Paper 2013
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Form 4 Chemistry Question Paper 2013
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Form 4 Chemistry Question Paper 2013
 - Setting Reading Goals Form 4 Chemistry Question Paper 2013
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Form 4 Chemistry Question Paper 2013
 - Fact-Checking eBook Content of Form 4 Chemistry Question Paper 2013
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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