

NAME \_\_\_\_\_ INDEX NUMBER \_\_\_\_\_

SCHOOL \_\_\_\_\_ DATE \_\_\_\_\_

## SIMPLE CLASSIFICATION OF SUBSTANCES

1. 1995 Q 15

Explain how you would separate a mixture of nitrogen and oxygen gases given that their boiling points are  $-196^{\circ}\text{C}$  and  $-183^{\circ}\text{C}$  respectively. (2 marks)

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.....

2. 1995 Q 8 P2

(a) The table below gives information about the major constituents of crude oil. Study it and answer the questions that follow.

| Constituent     | Boiling point ( $^{\circ}\text{C}$ ) |
|-----------------|--------------------------------------|
| Gases           | Below $-90$                          |
| Petrol          | $40-175$                             |
| Kerosene        | $175-250$                            |
| Diesel oil      | $250-350$                            |
| Lubricating oil | $350-400$                            |
| Bitumen         | Above $400$                          |

(i) Which one of the constituents of crude oil has molecules with the highest number of carbon atoms? (2 marks)

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(ii) Name the process you would use to separate a mixture of petrol and diesel and explain how the separation takes place. (2 marks)

.....

.....

(iii) Explain why the constituent of crude oil and write its formula. (1 mark)

# Form 1 Chemistry Questions

**R Sandford**

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