

Limiting and Excess Reagents

in this thought experiment we are considering a chemical reaction of the form:

1A + B

→ C

At a certain time some amount of each reactant (say x moles of A and y moles of B) are mixed and the reaction proceeds. In this case there is some limiting reagent (the one that is used up first) and some excess reagent (the one that is left over). The limiting reagent is the one that is used up first. The excess reagent is the one that is left over. The limiting reagent is the one that is used up first. The excess reagent is the one that is left over.

Model 1 - Assembling a Race Car



1. How many of each component are needed to assemble 1 complete car?

Chassis (10)

Engine (10)

Wheel (10)

Car (10)

2. How many of each component are needed to assemble 1 complete car with 1000 parts?

Chassis (1000)

Engine (1000)

Wheel (1000)

Car (1000)

10 x 1000 parts

10 x 1000 parts

10 x 1000 parts

10 x 1000 parts

3. Assuming the production of 1 complete car, how many of the remaining parts are left over?

1000 Chassis, 1000 Engine, 1000 Wheel, 1000 Car

Chassis: $\frac{1000}{10} = 100$

Engine: $\frac{1000}{10} = 100$

4. How many complete cars can be assembled from 1000 parts of each component?

Chassis: $\frac{1000}{10} = 100$

Engine: $\frac{1000}{10} = 100$

Wheel: $\frac{1000}{10} = 100$

Flinn Scientific Pogil Limiting And Excess Reactants

Lifelique

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

Flinn Scientific Pogil Limiting And Excess Reactants:

Limiting Reactant and Percent Yield, 2019 This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant calculate the theoretical yield of a reaction and calculating the percent yield of a reaction

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