

SECTION - A

Q. 1. Select and write the correct answer for the following multiple choice type of questions :

[10]

(i) The product obtained in the following reaction

P 239
11.2



- (a) Pent-3-en-1-ol (b) ☒ Pentan-1-ol
(c) Pentan-2-ol (d) Pentanal

(ii) Amongst the following, the solubility of which ionic solid decreases with increase in temperature?

P 30
11)

- (a) KNO_3 (b) NaBr
(c) ☒ Na_2SO_4 (d) KCl

(iii) The correct IUPAC name of $\text{Na}_3 [\text{AlF}_6]$ is

P 196
Table
9.3

- (a) ☒ Sodium hexafluoroaluminate (III)
(b) Sodium hexafluoroaluminate (II)
(c) Sodium hexafluoroaluminium (III)
(d) Sodium hexafluoroaluminium (II)

(iv) Which of the following acids has highest pKa value?

P 277
Table
12.9

- (a) Mono chloroacetic acid (b) Dichloroacetic acid
(c) Trichloroacetic acid (d) ☒ Acetic acid

(v) Number of carbon atoms present in isoprene unit is ____.

P 322
15.3

- (a) 6 (b) ☒ 5
(c) 4 (d) 3

(vi) The colourless transition metal ion amongst the following is

P 173
(a) ☒ Cu^+ 3d¹⁰
(c) Ni^{2+}

- (b) Cu^{2+}
(d) Co^{2+}

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DP Hallahan



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