

Results and Calculations:

→ Copper Salt:

• Calculations A:

	Trial 1	Trial 2	Trial 3
Mass of sample (g)	0.703 g	0.702 g	0.7011 g
Mass of filter paper (g)	1.092 g	1.100 g	1.0701 g
Mass of filter paper + Copper precipitate (g)	1.403 g	1.309 g	1.3040 g
Mass of Copper precipitate (g)	0.311 g	0.209 g	0.2335 g
Percent of Copper in original sample (%)	44.24 %	34.045 %	33.36 %

• Calculations B:

	Calculations
Average percent Cu in original sample (%)	$\frac{x_1 + x_2 + x_3}{3} = \frac{44.24 + 34.045 + 33.36}{3} = 37.20\%$
Standard deviation	$S = \sqrt{\frac{\sum (x_i - \bar{x})^2}{n - 1}} = \sqrt{\frac{7.84^2 + 3.16^2 + 3.84^2}{3 - 1}} = 4.895$
Relative Standard deviation	$\frac{S}{\text{mean}} = \frac{4.895}{37.20} = 0.1318$
Reported percent copper	mean \pm S = 37.20 \pm 4.895

• Observations:

Observations
The reason of adding water on the precipitate after filtration?
to get rid of excess HCl
The colour of the copper precipitate?
Blue

Gravimetric Determination Of Copper Lab Report

RD Boyd



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The Collegian and Progress of India ,1912 *ARS-72* United States. Agricultural Research Service,1965 *Nuclear Science Abstracts* ,1976 **Journal of the Society of Chemical Industry** ,1923 **Experiment Station Record** United States. Office of Experiment Stations,1896 Technical Abstract Bulletin Defense Documentation Center (U.S.),1961-10
Current Chemical Papers ,1956 A classified world list of new papers in pure chemistry *Report summaries* United States. Environmental Protection Agency,1983 **Chemical Abstracts** ,1909 Energy Research Abstracts ,1993
Indian Science Abstracts ,1983 *Iron Trade Review* ,1914 **Scientific and Technical Aerospace Reports** ,1995
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