

Solution (A) (150 μ M xylenol orange) (140 mM NaCl) (1.35M glycerol; pH 1.75)	A weight of 114mg of xylenol orange and 8.18g of NaCl were dissolved in 900 mL of 25 mM H ₂ SO ₄ solution, then 100 mL of glycerol was added to the solution. (This reagent is stable for at least 6 months at 4°C).
Solution (B) (5 mM ferrous ammonium sulfate) (10 mM O-dianisidine)	A weight of 1.96g of ferrous ammonium sulfate and 2.44g of O-dianisidine were dissolved in 1000ml of 25 mM H ₂ SO ₄ solution. (This reagent is stable for at least 6 months at 4°C). (Warning): Gloves and face mask must be wearing when dealing with o-dianisidine because it is a toxic and carcinogenic substance.
Solution (C) Standard Solution H ₂ O ₂ (200 μ M)	This solution was prepared by diluting (13 μ L) of (15%) commercial hydrogen peroxide (485 \times 10 ⁴ μ M) up to (1 L) with distilled water.

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... Merck & Co,1898 Manual of the Chemical Analysis of Rocks Henry Stephens Washington,1919 *University of Michigan Official Publication* ,1947 *Register of the University of California* University of California (1868-1952),1941

Catalogue of the University of Michigan University of Michigan,1958 Announcements for the following year included in some vols *General Register* University of Michigan,1942 Announcements for the following year included in some vols

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