1. What volume of hydrogen gas at STP is produced from the reaction of 50.0g of Mg and 75.0 grams of HCl? How much of the excess reagent is left over (in grams)?

$$Mg(s) + 2 HCl(aq) \rightarrow MgCl_2(s) + H_2(g)$$

Do a standard mass to mass problem starting with each reactant

50.0 gMg	1mol Mg	1mol H ₂	22.4 L	=	46.1 L H ₂
	24.31g Mg	1mol Mg	1mol H ₂		

HCl is the limiting reactant!!

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K Payea

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Hspi The Pogil Project Limiting Reactants Ciy Vm Answers Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has be more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Hspi The Pogil Project Limiting Reactants Ciy Vm Answers**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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