

Density Worksheet

Physical Science

$D=m/V$

Densities of Common Substances @ 20°C			
Substance	Density (g/cm ³)	Substance	Density (g/cm ³)
Oxygen	0.00133	Aluminum	2.70
Hydrogen	0.000084	Iron	7.87
Ethanol	0.785	Copper	8.96
Benzene	0.880	Silver	10.5
Water	1.000	Lead	11.34
Magnesium	1.74	Mercury	13.6
Salt (sodium chloride)	2.16	Gold	19.32

1. The ratio of an object's mass to its _____ is called the *density* of the object.
2. A kilogram of lead occupies a much smaller volume than a kilogram of water, because _____ has a much higher *density*.
3. For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
 - a. mass = 453 g; volume = 225 cm³
 - b. mass = 5.0 g; volume = 10.0 cm³
 - c. mass = 26.1 g; volume = 2.0 mL
4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
5. A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

Gravity And Acceleration Physical Science If8767

Answers

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web a economic system b social system c cultural attitudes d political system view answer is not an indian multinational company a unilever b asian paints c piramal d wipro view answer which of the following is not a force in the porter five forces model a buyers b suppliers c complementary products d industry rivalry

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