Grade	12 Paper 2 (Chemistry)	MEMO	September 2014	
1.1	B			
1.2	A			
1.4	8			
1.5	A			
1.6	A -			
1.7	D **			
1.8	D			
1.9	c			
1.10	D ·		$(2 \times 10 = 20)$	
2.1	H - C - C - C - C - C - C - C - C - C -	—	-H (3)	
2.2	They are volatile and evaporate • quickly.		(1)	
2.4	hydrolysis/elimination/esterification/conden	302,524 / 90235		
2.4	hydrolysis/elimination/esterification/conden	sation (any	one) (1)	
2.5	heat * and sulphuric acid (H <sub>2</sub> SO <sub>4</sub> ) *		(2)	
3.1	Methane is a non-polar substance  → and is the and contains and –OH group → which allows			
3.2	Methane and Methanal		(2)	
3.3	As the mass of the carboxylic acid increases, the boiling point of the carboxylic acids also increases. This is because larger molecules have more intermolecular forces. than smaller molecules.			
4.1.1	Hydration		(1)	
4.1.2	No <sup>✓</sup> because it needs a catalyst and heat <sup>✓</sup>	to take place.	(2)	
4.1.3	propan ~ -1-ol ~		(2)	
4.1.4	primary *		(1)	
4.1.5	No ♥, also propan-2-ol ♥		(2)	
4.2.1	hydrohalogenation		(1)	
4.2.2	no water present		(1)	

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Albert J. Churella

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## **Table of Contents Grade Memo September 2014**

- 1. Understanding the eBook Grade Memo September 2014
  - The Rise of Digital Reading Grade Memo September 2014
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Grade Memo September 2014
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Grade Memo September 2014
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Grade Memo September 2014
  - Personalized Recommendations
  - Grade Memo September 2014 User Reviews and Ratings

- Grade Memo September 2014 and Bestseller Lists
- 5. Accessing Grade Memo September 2014 Free and Paid eBooks
  - o Grade Memo September 2014 Public Domain eBooks
  - Grade Memo September 2014 eBook Subscription Services
  - Grade Memo September 2014 Budget-Friendly Options
- 6. Navigating Grade Memo September 2014 eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - Grade Memo September 2014 Compatibility with Devices
  - Grade Memo September 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Grade Memo September 2014
  - Highlighting and Note-Taking Grade Memo September 2014
  - Interactive Elements Grade Memo September 2014
- 8. Staying Engaged with Grade Memo September 2014
  - o Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Grade Memo September 2014
- 9. Balancing eBooks and Physical Books Grade Memo September 2014
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Grade Memo September 2014
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Grade Memo September 2014
  - Setting Reading Goals Grade Memo September 2014
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade Memo September 2014
  - Fact-Checking eBook Content of Grade Memo September 2014
  - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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