

# Physical Science Curriculum

## *Unit Sequence*

SEMESTER ONE	SEMESTER TWO
<p><b>Unit 1: Scientific Method</b></p> <ul style="list-style-type: none"> <li>• Lab Basics: Safety and equipment</li> <li>• Measurement</li> <li>• Dimensional analysis and scientific notation</li> <li>• Scientific method</li> </ul> <p><b>Unit 2: Motion and Force</b></p> <ul style="list-style-type: none"> <li>• Describing motion</li> <li>• Acceleration</li> <li>• Newton's laws</li> </ul> <p><b>Unit 3: Energy</b></p> <ul style="list-style-type: none"> <li>• Nature of energy</li> <li>• Conservation of energy</li> <li>• Thermal energy</li> <li>• Work and power</li> </ul> <p><b>Unit 4: Electricity and Magnetism</b></p> <ul style="list-style-type: none"> <li>• Charge</li> <li>• Current</li> <li>• Circuits</li> <li>• Magnets</li> </ul> <p><b>Unit 5: Waves</b></p> <ul style="list-style-type: none"> <li>• Nature and properties of waves</li> <li>• Behaviors of waves</li> <li>• Electromagnetic spectrum</li> </ul>	<p><b>Unit 6: Matter</b></p> <ul style="list-style-type: none"> <li>• Classification of matter</li> <li>• Properties of matter</li> <li>• States of matter</li> <li>• Solutions</li> </ul> <p><b>Unit 7: Atomic Structure</b></p> <ul style="list-style-type: none"> <li>• Structure of the atom</li> <li>• Isotopes</li> <li>• Periodic table of elements</li> </ul> <p><b>Unit 8: Bonding</b></p> <ul style="list-style-type: none"> <li>• Stability and bonding</li> <li>• Naming ionic compounds</li> <li>• Naming covalent compounds</li> </ul> <p><b>Unit 9: Reactions</b></p> <ul style="list-style-type: none"> <li>• Balancing reactions</li> <li>• Classifying reactions</li> <li>• Energy in reactions</li> <li>• Acids and bases <i>(if time)</i></li> <li>• Nuclear chemistry <i>(if time)</i></li> </ul>

# Grade 1scope Of Physical Science Paper 2014

**RM Cervero**



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