Name: _____

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Balancing Chemical Equations

Balance the following chemical equations.

1. 3 KOH + 1
$$H_3PO_4$$
 -> 1 K_3PO_4 + 3 H_2O

2.
$$4$$
 NH₃ + 5 O₂ \longrightarrow 4 NO + 6 H₂O

4.
$$\underline{2}$$
 CaS + $\underline{2}$ H₂O \longrightarrow $\underline{1}$ Ca(HS)₂ + $\underline{1}$ Ca(OH)₂

5.
$$\underline{1}$$
 Au₂S₃ + $\underline{3}$ H₂ \longrightarrow $\underline{2}$ Au + $\underline{3}$ H₂S

7.
$$\underline{1}$$
 Ba₃N₂ + $\underline{6}$ HF \longrightarrow $\underline{3}$ BaF₂ + $\underline{2}$ NH₃

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