CASCALLER IN PROPERTY.

(There is no Prelaboratory Assignment for this exercise. Make entries in your lab notebook for anything that is an observation; for your report, turn in this page and the following page.)

Making molecular models

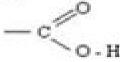
Make models of the following functional groups (additional red O balls will be needed):

make one C-C group

make four -H

make two C=C groups

make four



groups

Make models of the following dicarboxylic acids and observe:

- A. Rotation around chemical bonds:
 - Make succinic acid, HOOC-CH2-CH2-COOH.

Does the molecule rotate around the central C-C bond?

2.Replace the central CH2-CH2 group with a HC=CH group (keep one COOH group on each end).

Does the molecule rotate around the central C=C bond?

- 3. What is the molecular formula (C_iH_jO_k)of your new molecule?
- B. Comparison of maleic and fumaric acids.

Maleic acid and fumaric acid have the same molecular formula (C₄H₄O₄) and the same condensed structural formula (HOOC-CH=CH-COOH). Your model is either maleic or fumaric acid.

- Compare your model with others in the class. What differences do you see?
- 2. Make a second model of the other form.

Geometric Isomers Lab Answers

Lei Shi

Geometric Isomers Lab Answers:

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