

Ted's shelf is 0.86m long. He has arranged 3 different toys below on the shelf. There is a gap of 15mm at both ends of the shelf and between each toy. Which of the toys has Ted displayed on the shelf? Write the name of the toys in alphabetical order.



Check

Hint: One of the three toys is 20cm shorter than the combined length of the other two toys.

Financial Maths Unit And Resources Year 5

JL Elias



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Jacaranda Maths Quest 12 General Mathematics Units 3 & 4 for Queensland, 2e learnON and Print Pauline Holland, Mark Barnes, 2025-11-24 **Resources in Education** ,1996 *International Practices to Promote Budget Literacy* Harika Masud, Helene Pfeil, Sanjay Agarwal, Alfredo Gonzalez Briseno, 2017-08-09 Budget literacy is defined as the ability to read decipher and understand public budgets to enable and enhance meaningful citizen participation in the budget process It is comprised of two main parts i a technical understanding of public budgets including familiarity with government spending tax rates and public debt and ii the ability to engage in the budget process comprising of practical knowledge on day to day issues as well as an elementary understanding of the economic social and political implications of budget policies the stakeholders involved and when and how to provide inputs during the annual budget cycle Given that no international standards or guidelines have been established for budget literacy education to date this book seeks to address this gap by taking stock of illustrative initiatives promoting budget literacy for youth in selected countries The underlying presumption is that when supply side actors in the budget process governments simplify and disseminate budget information for demand side actors citizens this information will then be used by citizens to provide feedback on the budget However since citizens are often insufficiently informed about public budgets to constructively participate in budget processes one way to empower them and to remedy the problem of budget illiteracy is to provide budget literacy education in schools to youth helping them evolve into civic minded adults with the essential knowledge needed for analyzing their government's fiscal policy objectives and measures and the confidence and sense of social responsibility to participate in the oversight of public resources This book elaborates on approaches learning outcomes pedagogical strategies and assessment approaches for budget literacy education and presents lessons that are relevant for the development improvement or scaling up of budget literacy initiatives

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$$2\text{Cu} + 2\text{AgNO}_3 \rightarrow 2\text{CuNO}_3 + 2\text{Ag}$$

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$$3\text{CO} + 7\text{H}_2 \rightarrow \text{C}_3\text{H}_8 + 3\text{H}_2\text{O}$$

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web start with 8 g of H₂S and multiply it with 2 mole H₂O over 2 mole H₂S from the equation

$$8\text{ g H}_2\text{S} \times \frac{2\text{ mole H}_2\text{O}}{2\text{ mole H}_2\text{S}} \times \frac{18\text{ g H}_2\text{O}}{1\text{ mole H}_2\text{O}} = 144\text{ g H}_2\text{O}$$

144 g H₂O is your theoretical yield so percent yield = $\frac{137\text{ g H}_2\text{O}}{144\text{ g H}_2\text{O}} \times 100 = 95\%$

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$$\text{C}_3\text{H}_6 + 3\text{O}_2 \rightarrow 3\text{CO} + 3\text{H}_2\text{O}$$

if the reaction of 21 g of O₂ produces a 47% yield how many grams of CO would be produced step one figure out how many moles of the limiting reagent you have

$$21\text{ g O}_2 \times \frac{1\text{ mole O}_2}{32\text{ g O}_2} = 0.66\text{ moles O}_2$$

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web a write a balanced equation for the reaction

$$4\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$$

b determine the theoretical yield of Al₂O₃

$$160\text{ g Al} \times \frac{1\text{ mole Al}}{27\text{ g Al}} \times \frac{2\text{ mole Al}_2\text{O}_3}{4\text{ mole Al}} \times \frac{101\text{ g Al}_2\text{O}_3}{1\text{ mole Al}_2\text{O}_3} = 160\text{ g Al}_2\text{O}_3$$

c determine the percent yield

$$\frac{\text{actual yield}}{\text{theoretical yield}} \times 100 = \frac{126\text{ g}}{160\text{ g}} \times 100 = 79\%$$

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web 7 na 2 c 2 o 4 is the limiting reactant percent yield 86 6 8 only four molecules can be made 9 this amount cannot be weighted by ordinary balances and is worthless 10 nitrogen is the limiting reagent 11 yes methane is the limiting reagent 12 c is the limiting reagent 4 33 g of h 2 are left over 13

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