



Section - Multiple Choice

1. Which of the following pairs forms a conjugate acid-base pair?

Acid	Base	Acid conjugate base	Base conjugate acid
H_2SO_4	HSO_4^-	HSO_4^-	H_2SO_4
H_2SO_4	SO_4^{2-}	HSO_4^-	HSO_4^-
HSO_4^-	SO_4^{2-}	HSO_4^-	HSO_4^-
HSO_4^-	H_2SO_4	HSO_4^-	HSO_4^-

- (A) H_2SO_4
 (B) HSO_4^-
 (C) SO_4^{2-}
 (D) H_2SO_4 and HSO_4^-

2. Which of the following is a conjugate acid-base pair of a weak base and a strong acid? The conjugate base of a strong acid is a weak base.

- (A) H_2SO_4 and HSO_4^- (strong acid and weak base)
 (B) HSO_4^- and SO_4^{2-} (weak acid and weak base)
 (C) HSO_4^- and SO_4^{2-} (strong acid and weak base)
 (D) HSO_4^- and SO_4^{2-} (weak acid and weak base)

3. Which of the following is a conjugate acid-base pair of a weak base and a strong acid?



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