

AS-Chapter 3 Handout

1. (a) Sodium reacts with cold water

(b) What would you see as the reaction proceeds?

.....

.....

.....

.....

(3)

(c) Write the balanced chemical equation for this reaction

.....

(1)

(d) Calculate the volume of gas produced if 3.0 g of sodium reacts with an excess of water

(One mole of any gas at the temperature and pressure of the experiment occupies 24 dm^3)

(3)
Total 7 marks

2. (a) (i) What is meant by the mass number of an atom?

.....

(1)

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Table of Contents June 2013 Chemistr Unit 3b Paper Ial

1. Understanding the eBook June 2013 Chemistr Unit 3b Paper Ial
 - The Rise of Digital Reading June 2013 Chemistr Unit 3b Paper Ial
 - Advantages of eBooks Over Traditional Books
2. Identifying June 2013 Chemistr Unit 3b Paper Ial
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an June 2013 Chemistr Unit 3b Paper Ial
 - User-Friendly Interface
4. Exploring eBook Recommendations from June 2013 Chemistr Unit 3b Paper Ial
 - Personalized Recommendations
 - June 2013 Chemistr Unit 3b Paper Ial User Reviews and Ratings
 - June 2013 Chemistr Unit 3b Paper Ial and Bestseller Lists
5. Accessing June 2013 Chemistr Unit 3b Paper Ial Free and Paid eBooks
 - June 2013 Chemistr Unit 3b Paper Ial Public Domain eBooks
 - June 2013 Chemistr Unit 3b Paper Ial eBook Subscription Services
 - June 2013 Chemistr Unit 3b Paper Ial Budget-Friendly Options
6. Navigating June 2013 Chemistr Unit 3b Paper Ial eBook Formats
 - ePub, PDF, MOBI, and More
 - June 2013 Chemistr Unit 3b Paper Ial Compatibility with Devices
 - June 2013 Chemistr Unit 3b Paper Ial Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of June 2013 Chemistr Unit 3b Paper Ial
 - Highlighting and Note-Taking June 2013 Chemistr Unit 3b Paper Ial
 - Interactive Elements June 2013 Chemistr Unit 3b Paper Ial
8. Staying Engaged with June 2013 Chemistr Unit 3b Paper Ial

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers June 2013 Chemistr Unit 3b Paper Ial
- 9. Balancing eBooks and Physical Books June 2013 Chemistr Unit 3b Paper Ial
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection June 2013 Chemistr Unit 3b Paper Ial
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine June 2013 Chemistr Unit 3b Paper Ial
 - Setting Reading Goals June 2013 Chemistr Unit 3b Paper Ial
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of June 2013 Chemistr Unit 3b Paper Ial
 - Fact-Checking eBook Content of June 2013 Chemistr Unit 3b Paper Ial
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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