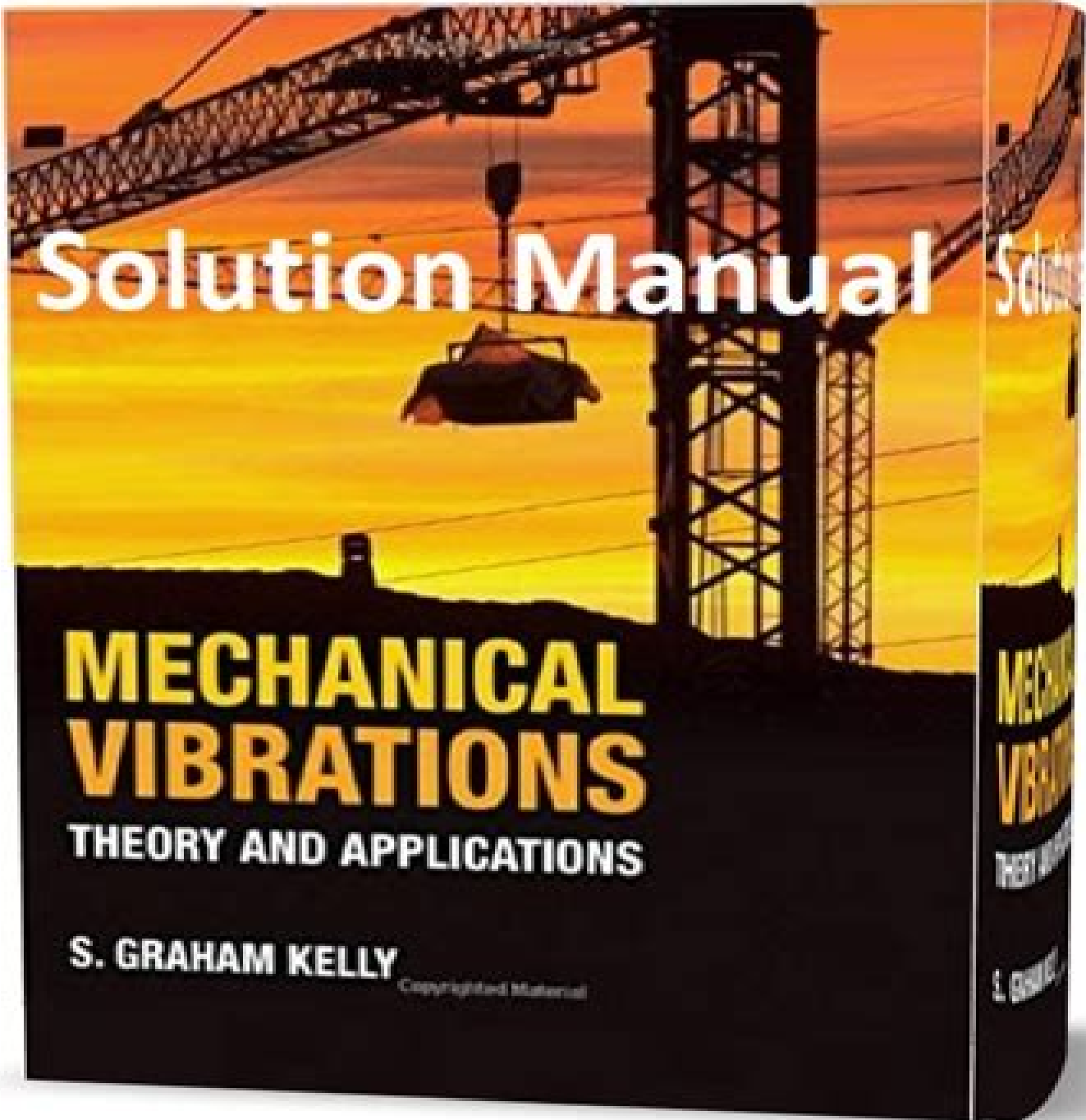




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THEORY AND APPLICATIONS

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web we utilize the equation that relate amount remaining initial mass and number of half lives $n = \frac{t}{T_{1/2}} = \frac{125 \text{ g}}{50 \text{ g}} = 2.5$ what is the half life of an isotope that is 80 remained after 16 days solution remaining 80 100 therefore $n = \frac{t}{T_{1/2}} = \frac{16 \text{ days}}{6.4 \text{ days}} = 2.5$ now using the half life equation in example 1 we have

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