

6. Polystyrene	Styrene	$\text{CH}_2=\text{CH}-\text{C}_6\text{H}_5$
7. Buna rubber	Buta-1,3-diene	$\text{CH}_2=\text{CH}-\text{CH}=\text{CH}_2$
8. Dacron (Terylene)	Ethane-1,2-diol + 1,4-benzenedioic acid	CH_2-CH_2 $\text{OH} \quad \text{OH}$ + $\text{HO}-\text{C}(=\text{O})-(\text{C}_6\text{H}_4)-\text{C}(=\text{O})-\text{OH}$
9. Nylon 6,6	1,6-diaminohexane + 1,6-hexanedioic acid	$\text{HO}-\text{C}(=\text{O})-(\text{C}_6\text{H}_4)-\text{C}(=\text{O})-\text{OH}$ + $\text{NH}_2-(\text{CH}_2)_6-\text{NH}_2$
10. Bakelite	Phenol + formaldehyde	$\text{HO}-\text{C}(=\text{O})-(\text{CH}_2)_4-\text{C}(=\text{O})-\text{OH}$ + $\text{C}_6\text{H}_5-\text{OH}$ + HCHO

2. On the basis of monomers.

a. Homopolymers

→ It is formed by only one type of monomer unit.

For eg:- polythene, polyvinyl chloride (PVC), Orlon, Teflon etc.

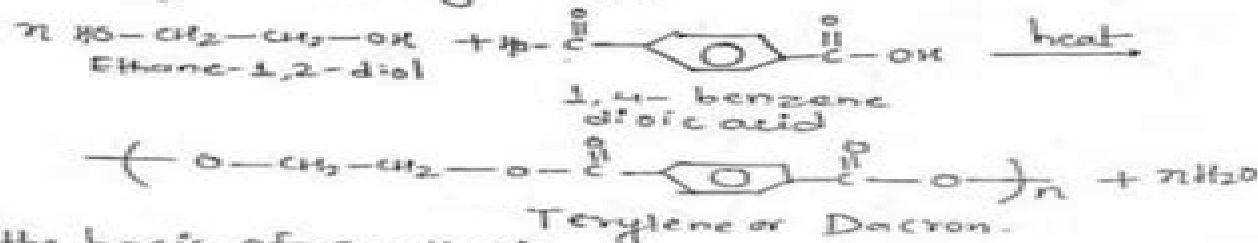


b. Copolymers:

→ It is formed by more than one type of monomers

→ Monomers are arranged in definite structural pattern in a random manner.

For eg Dacron, Nylon 6,6



3 On the basis of synthesis.

a. Addition polymers

- Large number of monomer molecules are added together to form high molecular weight.

For eg:- pvc, orlon, te flon

Mcq In Nuclear Chemistry

Kathleen Armour



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