

1.	(a)	(i)	electrons ...16.... protons16 neutrons ...16.... All correct (1)	1	[8]
		(ii)	they are isotopes (1) The word 'isotope' or some derivative is essential	1	
		(iii)	S^{2-} / sulphide(2-) ion / sulphur -2 ion / sulphur with two extra electrons (1) but not just S or sulphur	1	
	(b)	(i)	$X(g) + e^- \rightarrow X^-(g)$ ($S(g) + e^- \rightarrow S^-(g)$) (1) $X(g) + e^- \rightarrow X^{2-}(g)$ ($S(g) + e^- \rightarrow S^{2-}(g)$) (1) can have Z in place of X ²⁻ but not Z ²⁻ Penalize omission of state symbol (g) only once.	2	
		(ii)	adding a negative onto a <u>negative species</u> requires energy to overcome repulsion (1) or second electron repelled by negative charge Not second electron repelled by first	1	
		(c)	(Bdecay from) Y produces only one isotope(^{39}Cl) (1) natural sample contains two isotopes (35 and 37) (1)	2	
	2.	(i)	magnesium / lead / iron (1)	1	
		(ii)	iron (1)	1	
		(iii)	carbon / phosphorus (1)	1	
		(iv)	magnesium / lead (1)	1	
		(v)	carbon / lead (1)	1	
		(b)	Helium (1) because it is the smallest atom (1) with an unshielded nucleus (1) Hydrogen can score max 1 for 'unshielded nucleus' If candidate chooses Carbon, i.e. uses list from part (a), can score 1 mark for reference to 'smallest atom'	3	
		(c)	(i) ability of a positive ion / cation to attract electrons / distort / pull the electron cloud (1) around an anion (1)	2	
		(ii)	size (1) Not atomic radius charge (1) charge density worth (2) volume of ion can score in place of size	2	
		(iii)	magnesium chloride magnesium ion smaller than calcium ion (1) polarizes / distorts / attracts electrons around chloride more (1)	2	

[14]

June 2013 Chemistry Unit 5 Mark Scheme

CL Gary



June 2013 Chemistry Unit 5 Mark Scheme:

The book delves into June 2013 Chemistry Unit 5 Mark Scheme. June 2013 Chemistry Unit 5 Mark Scheme is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into June 2013 Chemistry Unit 5 Mark Scheme, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to June 2013 Chemistry Unit 5 Mark Scheme
- Chapter 2: Essential Elements of June 2013 Chemistry Unit 5 Mark Scheme
- Chapter 3: June 2013 Chemistry Unit 5 Mark Scheme in Everyday Life
- Chapter 4: June 2013 Chemistry Unit 5 Mark Scheme in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of June 2013 Chemistry Unit 5 Mark Scheme. The first chapter will explore what June 2013 Chemistry Unit 5 Mark Scheme is, why June 2013 Chemistry Unit 5 Mark Scheme is vital, and how to effectively learn about June 2013 Chemistry Unit 5 Mark Scheme.

3. In chapter 2, this book will delve into the foundational concepts of June 2013 Chemistry Unit 5 Mark Scheme. The second chapter will elucidate the essential principles that must be understood to grasp June 2013 Chemistry Unit 5 Mark Scheme in its entirety.

4. In chapter 3, the author will examine the practical applications of June 2013 Chemistry Unit 5 Mark Scheme in daily life. This chapter will showcase real-world examples of how June 2013 Chemistry Unit 5 Mark Scheme can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of June 2013 Chemistry Unit 5 Mark Scheme in specific contexts. This chapter will explore how June 2013 Chemistry Unit 5 Mark Scheme is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about June 2013 Chemistry Unit 5 Mark Scheme. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of June 2013 Chemistry Unit 5 Mark Scheme.

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