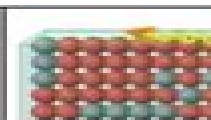


Activity B:
Measuring
half-life

Get the Gizmo ready:

- Click **Reset**.
- Select **Isotope A** from the left drop-down menu.
- Check that **Theoretical decay** is selected.



Introduction: Different **isotopes** of the same element have the same number of protons but different numbers of **neutrons** in the nucleus. Some isotopes are radioactive.

Question: How do we find the half-life of a radioactive isotope?

1. Observe: Select the **GRAPH** tab, and click **Play**. Based on the graph, what is your estimate of the half-life of isotope A?

15

2. Measure: Turn on the **Half-life probe**. Use the probe to measure how long it takes for exactly one-half of the original radioactive atoms to decay.

What is the exact half-life of isotope A?

31

3. Collect data: In the first row of the table below, write how many seconds represent one half-life, two half-lives, and so forth. On the next row, predict the number of radioactive atoms that will be present at each time. Then use the probe to find the actual values.

| Half-life | 0 | 1 | 2 | 3 | 4 | 5 |
|-------------------------------|-----|----|----|----|-----|-----|
| Time (seconds) | | 31 | 62 | 94 | 125 | 155 |
| Predicted # radioactive atoms | 128 | 64 | 32 | 16 | 8 | 4 |
| Actual # radioactive atoms | 128 | 64 | 32 | 16 | 8 | 4 |

4. Calculate: Calculate the percentage of radioactive atoms that are left after each half-life.

| Half-life | 0 | 1 | 2 | 3 | 4 | 5 |
|------------------------------|-----|----|----|------|------|-------|
| Percentage radioactive atoms | 100 | 50 | 25 | 12.5 | 6.25 | 3.125 |

5. Apply: Suppose you found a material in which 12.5% of the original radioactive atoms were present. If the half-life is 47 years, how old is the material?

around 200 years of age

Half Life Lab Answer Key

WJ Hussar

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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